

# Analysis of Underground Storage Tank System Materials to Increased Leak Potential Associated with E15 Fuel

July 2012

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**ANALYSIS OF UNDERGROUND STORAGE TANK  
SYSTEM MATERIALS TO INCREASED LEAK  
POTENTIAL ASSOCIATED WITH E15 FUEL**

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## CONTENTS

	Page
LIST OF FIGURES .....	v
LIST OF TABLES .....	vii
ACRONYMS .....	ix
FOREWARD .....	xi
ACKNOWLEDGMENTS .....	xiii
EXECUTIVE SUMMARY .....	xv
1. INTRODUCTION .....	1
1.1 HISTORY AND BACKGROUND .....	1
1.2 ETHANOL COMPATIBILITY AND SOLUBILITY .....	2
1.3 FUELING DISPENSER MATERIALS AND ETHANOL COMPATIBILITY STUDY .....	5
2. UNDERGROUND TANKS & PIPING SYSTEMS .....	6
2.1 METALLIC MATERIALS FOR TANKS, COMPONENTS, AND PIPING SYSTEMS .....	7
2.2 POLYMER PIPING AND TANK SYSTEMS .....	8
2.2.1 Flexible Plastic Piping .....	8
2.2.2 Fiber-reinforced Plastic Tanks & Piping .....	13
3. ELASTOMERS, SEALANTS, COUPLINGS AND FITTINGS .....	19
3.1 ELASTOMERS .....	19
3.2 PIPE THREAD SEALANTS .....	21
3.3 COUPLINGS AND FITTINGS .....	22
4. CONCLUSIONS .....	24
4.1 CONCLUSION ON TANKS AND PIPING MATERIALS .....	24
4.2 CONCLUSION ON ELASTOMERS, SEALANTS, COUPLINGS AND FITTINGS .....	25
5. REFERENCES .....	25



## LIST OF FIGURES

Figure		Page
1	Electrical conductivity of gasoline as a function of ethanol concentration. ....	3
2	Electrical conductivity of gasoline as a function of ethanol and water content. ....	4
3	Total Hansen Solubility Parameter as a function of ethanol concentration.....	5
4	Cross-section diagram of flexible piping showing an example of the layering position and arrangement of materials used in double-walled designs. ....	9
5	Volume swell results for representative barrier materials used in underground piping. ....	10
6	Absolute hardness results for representative barrier materials used in flexible piping. ....	11
7	Point change in hardness (from baseline) for representative barrier materials used in flexible piping.....	11
8	Diagram of fiber-reinforced plastic piping.....	14
9	Volume swell results for UST resins following exposure to Fuel C and CE25a. ....	15
10	Mass change for UST resins following exposure to Fuel C and CE25a. ....	15
11	Point change in hardness for the UST resin samples following exposure to Fuel C and CE25a. ....	17
12	Photograph showing the Batch 1 specimens before and after exposure to Fuel C, CE50a, and CE85a. ....	18
13	Photograph showing the Batch 2 specimens before and after exposure to Fuel C, CE50a, and CE85a. ....	18
14	Photograph showing the Batch 3 specimens before and after exposure to Fuel C, CE50a, and CE85a. ....	19
15	Volume swell and point change in hardness for elastomers exposed to Fuel C, CE10a, CE17a and CE25a. ....	20
16	Percent mass change and point change in hardness for elastomers exposed to Fuel C, CE10a, CE17a and CE25a following dry-out. ....	21
17	Simplified schematic showing attachment of a coupling to flexible plastic piping. ....	22
18	Schematic diagram of the butt and strap joint. ....	23
19	Schematic showing a common arrangement of using adhesive on FRP piping. ....	23





## LIST OF TABLES

Tables	Page
1	List of materials evaluated in intermediate ethanol blends compatibility study. (Materials identified as being used in UST systems are highlighted.) ..... 6
2	Breakdown of piping materials. <sup>5,19</sup> ..... 7
3	Flexible piping materials according to manufacturer. .... 8
4	Volume swell results for representative barrier materials used in flexible underground piping..... 13
5	Measured and calculated results for PVDF and Nylon 6 ..... 13
6	Measured and calculated results for UST resins Vipel F774 and Vipel F085..... 16



## ACRONYMS

ASTM	American Society for Testing and Materials
DOE	Department of Energy
E10	Gasoline containing 10% ethanol by volume
E15	Gasoline containing 15% ethanol by volume
E50	Gasoline containing 50% ethanol by volume
E85	Gasoline containing 85% ethanol by volume
EISA	Energy Independence and Security Act
EPA	U. S. Environmental Protection Agency
F-HDPE	Fluorinated high density polyethylene
Fuel C	A gasoline representative test fuel composed of 50% vol. toluene and 50% vol. isooctane
FRP	Fiber-reinforced plastic
FFV	Flex-Fuel Vehicle
HDPE	High density polyethylene
HSP	Hansen Solubility Parameter
ISO	International Organization for Standardization
LG	Leaded gasoline
MIC	Microbial-induced corrosion
MTBE	Methyl tertiary butyl ether
NBR	Acrylonitrile (or nitrile) butadiene rubber
NREL	National Renewable Energy Laboratory
ORNL	Oak Ridge National Laboratory
PBT	polybutylene terephthalate
PEI	Petroleum Equipment Institute
PET	polyethylene terephthalate
PP	polypropylene
PTFE	polytetrafluoroethylene
PCV	polyvinyl chloride
PVDF	polyvinylidene fluoride
RFS	Renewable Fuel Standard
S	Siemens (unit of electrical conductivity)
SAE	Society of Automotive Engineers
SBR	Styrene butadiene rubber
UL	Underwriters Laboratories
UST	Underground Storage Tank
VS	Volume swell
VTP	DOE Vehicle Technologies Program

DOE Office of Biomass Program



## **FOREWARD**

It is not the purpose of this report to define the acceptable limits of material performance or to rate individual materials. Rather, the purpose of this study was to assess critical property changes (volume, hardness, mass, etc.) for representative classes of materials used in underground storage tank systems with exposure to E15.



## **ACKNOWLEDGMENTS**

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## EXECUTIVE SUMMARY

### E.1 Background

The Energy Independence and Security Act (EISA) of 2007 was an omnibus energy policy law designed to move the United States toward greater energy security and independence.<sup>1</sup> A key provision of EISA modified the Renewable Fuel Standard (RFS) which requires the nation to increase the volume of renewable fuel blended into transportation fuels from 7.5 billion gallons by 2012 to 36 billion gallons by 2022. Ethanol is the most widely used renewable fuel, and increasing the ethanol content in gasoline to 15% offers a means of getting significantly closer to the 36 billion gallon goal. In March 2009, Growth Energy (a coalition of ethanol producers and supporters) requested a waiver from the United States Environmental Protection Agency (EPA) to allow the use of 15% ethanol in gasoline.<sup>2</sup> In response the US EPA granted two partial waivers that allow (but do not require) E15 in 2001 and newer light-duty vehicles. Prior to the waiver being granted, uncertainties arose as to whether the additional fuel ethanol (from 10% to 15%), would cause an increase in leaking of underground storage tank (UST) systems, which include not only the tank but also the piping and connecting hardware.

The USEPA Office of Underground Storage Tanks was interested in determining how many (of the nearly 600 thousand) federally regulated underground storage tank (UST) systems across the U.S. could have releases or other failures if the ethanol content in gasoline increases from 10 volume percent to 15 volume percent. To better assess the leak potential, the EPA commissioned a study at Oak Ridge National Laboratory to develop a means to determine the potential of changes in releases and other failures if E15 fuel is stored in UST systems. Part of this effort was to develop an approach to estimate likelihood of failures and approaches for mitigating consequences associated with these failures. Currently, the lack of availability of data is the most significant barrier that prevents EPA from being able to perform the analysis.

The initial approach was to develop and apply a probabilistic failure analysis tool based on expert elicitation to estimate how many more releases would occur if E15 replaced E10 in regulated UST systems. The key resources needed to establish this tool were opinions provided by industry and regulatory experts to quantify (most likely values and uncertainties) the critical variables that impact failure likelihood estimates. Unfortunately, over the course of the investigation, it was discovered that there was no information on the performance of existing UST systems with E15 and the state/industry experts were unable to speculate on E15's impact to UST systems. As a result, the project objective was redirected to address the added leak potential (or incompatibility) of UST system materials when switching from E10 to E15. The data used to make this assessment were obtained primarily from the ORNL intermediate blend compatibility study.<sup>3</sup> The ORNL study included metal and polymeric materials typically used in UST systems, and these materials were evaluated in aggressive test fuel formulations representing E0, E10, E15, and E25. Later studies investigated material compatibility to E50 and E85.

The elastomeric and metallic materials were exposed to Fuel C, CE10a, and CE17a test fuels, which are based on standard fluids described by the American Society for Testing and Materials (ASTM) and the Society of Automotive Engineers (SAE) for use in fuel-material compatibility studies. SAE Reference Fuel C (also known as Fuel C) is a 50:50 mix of isooctane and toluene, and was used as the base fuel in the ethanol-blended test fuels, where it is represented by the "C" nomenclature. The ethanol was made to an aggressive formulation per SAE J1681,<sup>4</sup> and is indicated by the letter "a". CE17a was chosen to represent E15 since fuel surveys have shown that the actual ethanol content in gasoline can vary by  $\pm 2\%$ . Plastic materials were only evaluated in Fuel C and CE25a. Therefore it was necessary to assess E10 and E15 performance through an interpolation process using the known solubility parameters for these materials and their performance in Fuel C and CE25a.

## **E.2 Experimental Approach**

The approach was to use the swell, mass change, and hardness data from the ORNL study to assess the risk of moving from E10 to E15. An extensive literature review was undertaken which was initially based on the EPA 22 state study<sup>5</sup> to accurately identify materials used in UST systems. The system components of interest included tanks, piping, sealants, and joined couplings. Piping was divided into three areas: metal, flexible plastic, and rigid fiberglass-reinforced plastic. Because most of the installed piping systems are plastic, these systems are discussed in greater detail. For the elastomeric and metallic materials, analysis was performed using results obtained from exposure to test fuels containing 10% and 17%. On the other hand, plastic materials were only exposed to Fuel C and CE25a. In order to estimate the level of swell (or solubility) for representative plastics in E10 and E15, an analysis was performed using the results obtained from the Fuel C and CE25a exposures and incorporating solubility theory. An estimate of the volume swell (at E10 and E15) was made by interpolating the results for Fuel C and CE25a.

## **E.3 Discussion and Analysis**

### ***Underground Storage Tanks and Piping Made of Steel***

For metal-based tanks and piping, corrosion via oxidation of the metal can directly lead to the creation of a leak. Another potential concern with higher ethanol content is the initiation of a new phase of corrosion, such that previously passivated areas (rust plugs) are attacked and removed, thereby leading to potential leaks. All metal USTs are composed of mild-carbon steel and around 98% of metal piping is also mild-carbon steel.<sup>5</sup> The other metal of interest is aluminum since aluminum parts are used on submersible turbine pumps, connections and dispenser nozzle. The ORNL intermediate-blend study included both steel and aluminum; the study showed negligible corrosion of either steel or aluminum immersed in either CE10a or CE17a.<sup>3</sup> However, the test conditions may not accurately reflect actual field situations, whereby the metal structure may be under stress or exposed to fuel that has become separated into two phases, one of which is aqueous. Both of these conditions (stress and exposure to aqueous liquid) are considered to be more conducive to corrosion. The specimens evaluated in the ORNL intermediate-blends study were not placed under stress, so the stress corrosion cracking potential of steel to either E10 or E15 cannot be ascertained.

Phase separation (of water) is another scenario that needs to be addressed. The level of water that can be dissolved into E15 is roughly twice the amount that can be dissolved in E10. Therefore, under identical conditions of phase separation (such as temperature excursions causing evaporation and condensation) E15 has the potential to generate twice the volume of aqueous phase than E10, which could translate to a higher corrosion (and therefore leak) potential. The presence of an aqueous phase is also a precondition for supporting microbial-induced corrosion (MIC), and if E15 has a higher potential for water formation, then MIC may also result in increased corrosion. If precautions are undertaken to keep water out of tanks, and stress corrosion cracking is not a factor, then the corrosion potential is minimized and E15 offers no added risk to metal corrosion than E10.

### ***Underground Storage Tanks and Piping Made of Fiberglass-reinforced Plastic (FRP)***

The other material used in the construction of USTs is fiberglass-reinforced plastic (FRP). FRP construction consists of initially placing an approximately 0.5mm thick layer of resin on a mandrel followed by adding an additional ~6mm layer of resin reinforced with fiberglass. The inner bare resin surface serves as the barrier layer to prevent fuel permeation and the fiberglass-reinforcement provides strength and elasticity. Some legacy designs also may incorporate a separate plastic film that was glued to the inside surface to provide a fuel-resistant barrier layer. The ORNL intermediate-blend materials

compatibility study<sup>3</sup> had evaluated four resin types representative of those used in legacy and modern FRP UST construction. One resin was used extensively prior to 1990 and therefore may not have been designed for E10 compatibility. Two of the test resins were introduced during the 1990s (post-1990), during which time E10 was beginning to be used in the marketplace. The fourth resin type was a new advanced resin developed for improved resistance to ethanol fuels. These four resins were made into test coupons (with no added fiberglass) and exposed to test fuels of Fuel C and CE25a.

Because E15 and E10 test fuels were not used in this evaluation, it was necessary to estimate resin performance in E10 and E15 using the swelling data obtained from the Fuel C and CE25a exposures. This estimation was performed by interpolating the measured swelling data using the differences in the known total Hansen Solubility Parameters (HSPs) for the resins and test fuels. (This procedure is described in detail in Section 2.1.1.) The solubility parameter is based on the free energy of mixing and is useful in predicting the mutual solubility (and therefore swell) between liquids and solid hydrocarbon materials. The pre-1990 resin was severely damaged from exposure to CE25a, along with one of the post-1990 resins. The remaining post-1990 resin and the advanced resin type both remained intact after exposure to CE25a, but they did swell to over 20% from their original volume with addition of ethanol. However, interpolation of these results using the Hansen Solubility Parameters suggests that the additional swell achieved from E10 to E15 will be around 1.5% (which is low). It is also important to keep in mind that the addition of fiberglass reinforcement to any of these resins will prevent significant swelling and debonding of the composite structure, since the fibers themselves do not swell.

The ORNL intermediate-blend materials compatibility study later included three legacy FRP UST specimens for evaluation, but they were only exposed to Fuel C, CE50a, and CE85a. One sample had a green coloration and contained a separate plastic barrier liner glued to the inner resin layer. The other two samples were amber in appearance, and of typical construction which consisted of an inner resin-only layer which was surrounded by a 6mm thick layer of fiberglass-reinforced resin. The resin used in the green UST survived Fuel C exposure but was severely degraded following exposure to CE50a and CE85a. In each case, the glue holding the plastic liner to the resin surface had dissolved, but, the plastic liner was unaffected. Unfortunately, the plastic composition of the liner was unknown, making it impossible to assess compatibility to E10 and E15. This particular UST design may be uncommon since, of the over two dozen samples provided to ORNL, it was the only one which had a separate inner liner and green resin. The other two USTs did not experience noticeable degradation or swell associated with exposure to the CE50a and CE85a test fuels. Because the difference in HSPs for resin and ethanol-blended gasoline increases with decreasing ethanol content, these epoxy resins should be more soluble in E50 and E85 than for intermediate E10 and E15 levels. Therefore, it is expected that USTs composed of amber resins will be compatible with gasoline containing 10 and 15 percent ethanol.

As of 2009, rigid FRP piping makes up around 58% of all installed piping systems.<sup>5</sup> The technology and materials used in the manufacture of FRP tanks also applies to underground FRP piping systems as well. Therefore the compatibility of FRP piping systems should be the same or similar to FRP underground storage tanks.

### ***Flexible Plastic Piping***

As of 2009, flexible plastic piping is estimated to make up around 13% of all installed piping systems,<sup>5</sup> but many new systems employ flexible plastic piping since these systems are easier to install. As a result, the percentage of flexible piping is expected to grow relative to other piping systems over the next 10 years. Typical compositional arrangement of most flexible piping includes an inner barrier liner with a layer of reinforcement (to provide strength) and an outer cover. Many of the outer layers are not compatible with ethanol and are only added to provide exterior protection and strength. The primary inner layer provides chemical resistance and a survey of flexible piping systems shows that the most common

inner permeation barrier material is polyvinylidene difluoride (PVDF). Other plastics used as permeation barriers are nylons and polyethylene terephthalate (PET). PVDF, PET, and several grades of nylon were evaluated in the ORNL intermediate-blends study along with the other plastic materials that were exposed to Fuel C and CE25a. As with the UST resins, the performance (volume swell) with exposure to E10 and E15 was estimated using the measured volume swelling for exposure to Fuel C and CE25a and the known HSPs for these materials. The resulting analysis indicates that flexible piping permeation barrier materials will not have added significant swell (less than 1%) when moving from E10 to E15. Therefore, the increase in risk associated with leaking when switching from E10 to E15 will be low.

### ***Elastomers, Sealants, Couplings and Fittings***

Couplings and fittings used to connect piping, the submersible turbine pump, and valves represent one of the highest potential locations for leaking in UST systems. There are two potential locations/sources of leaks associated with fittings. One is where the coupling attaches to the piping and the other one is at the fitting-to-fitting seal interface. In many (but not all) cases fluorocarbons are used as interfacial seals between fittings. Fluorocarbons have been shown to be compatible with ethanol and it is unlikely that a properly installed fluorocarbon elastomer will leak when exposed to either E10 or E15. For metal and some rigid FRP piping systems, pipe thread sealants may be employed to seal fittings via threaded attachments. Some legacy pipe thread sealants were shown to be incompatible with gasoline containing 10% aggressive ethanol and would clearly not be acceptable for E15 use either. Newer engineered products (such as fluoroelastomers) have been developed for ethanol-blended gasoline and these sealants have been shown to be compatible with gasoline containing up to 25% aggressive ethanol.

For flexible piping systems a stainless steel coupling is normally compression fitted to the outer surface of the pipe so the leak potential is very low for properly installed couplings. In contrast fittings attached to rigid FRP systems typically utilize an adhesive to maintain a seal between the coupling and the outer pipe wall. Adhesives designed for fuel ethanol use are available. This material type was not included in the ORNL intermediate-blend study and its performance in either E10 or E15 was not ascertained. For rigid FRP pipe-to-pipe joining, fiberglass reinforced resin is also frequently applied to the joined ends in a butt-and-wrap arrangement. Since the wrapping is composed of fiberglass-reinforced resin similar to the piping itself, the leak potential with exposure to E15 for a properly installed joint should be low since the increase of swell associated with E15 (relative to E10) is estimated to be small (1.5%). It is important to note that the joined sections have lower structural integrity (mechanical strength) than the pipe as a whole, but should not leak as a primary result of the fuel exposure.

## **E.4 Conclusions**

In general, the materials used in existing UST infrastructures would not be expected to exhibit compatibility concerns when moving from E10 to E15. The volume swell and hardness results of tested polymer materials were not significantly different when exposed to either CE10a or CE15a, although significant changes were observed when these fuels are compared to the E0 formulation. The indication is that UST systems were affected by switching from E0 to E10. However, since E10 and E15 produce similar results, compatibility is not expected to be altered noticeably when moving from E10 to E15. The metallic materials showed negligible corrosion as long as phase separation did not occur. If an aqueous phase is formed, then the possibility for aggressive corrosion exists. Therefore, the proper application of biocides and water monitoring is likely to be more critical at preventing corrosion for gasoline fuel containing ethanol.

# **1. INTRODUCTION**

## **1.1 HISTORY AND BACKGROUND**

In the United States oil dependence is driven primarily by the transportation sector. Transportation accounts for 69% of the total oil consumption in the United States, and the industry itself is around 90% oil dependent (and the remainder being natural gas, propane, electric and ethanol).<sup>6</sup> In 2008 the average daily oil consumption equivalent used the U.S. transportation sector was approximately 14 million barrels. This rate is projected to increase to around 16 million barrels per day by 2025.<sup>7</sup> Currently, the bulk of our oil usage is provided by other countries as foreign oil imports and makes up around 57% of the total oil usage.<sup>8</sup> This dependency impacts our nation's security, since our oil supply is determined partly by other countries, some of whom are not friendly to the United States. Foreign disruption has been shown to negatively impact the nation's economy and makes the U. S. more vulnerable during times of international crisis.

The Energy Independence and Security Act (EISA) of 2007 was enacted by Congress to move the nation toward increased energy independence by increasing the production of renewable fuels to meet its transportation energy needs. The law establishes a new renewable fuel standard (RFS) that requires the nation to use 36 billion gallons annually (2.3 million barrels per day) of renewable fuel in its vehicles by 2022. Ethanol is the most widely used renewable fuel in the United States, and its production has grown dramatically over the past decade. According to EISA and RFS, ethanol (produced from corn as well as cellulosic feedstocks) will make up the vast majority of the new renewable fuel requirements. However, ethanol use limited to E10 and E85 (in the case of flex fuel vehicles or FFVs) will not meet this target. Even if all of the E0 gasoline dispensers in the country were converted to E10, such sales would represent only about 15 billion gallons per year.<sup>9</sup> If 15% ethanol, rather than 10% were used, the potential would be up to 22 billion gallons. The vast majority of ethanol used in the United States is blended with gasoline to create E10, that is, gasoline with up to 10 % ethanol. The remaining ethanol is sold in the form of E85, a gasoline blend with as much as 85% ethanol that can only be used in FFVs. Although the U. S. Department of Energy (DOE) remains committed to expanding the E85 infrastructure, that market will not be able to absorb projected volumes of ethanol in the near term. Given this reality, DOE and others have begun assessing the viability of using intermediate ethanol blends as one way to transition to higher volumes of ethanol.

In October of 2010, the U. S. Environmental Protection Agency (EPA) granted a partial waiver to the Clean Air Act allowing the use of fuel that contains up to 15% ethanol for the model year 2007 and newer light-duty motor vehicles. This waiver represents the first of a number of actions that are needed to move toward the commercialization of E15 gasoline blends. On January 2011, this waiver was expanded to include model year 2001 light-duty vehicles, but specifically prohibited use in motorcycles and off-road vehicles and equipment.<sup>2</sup>

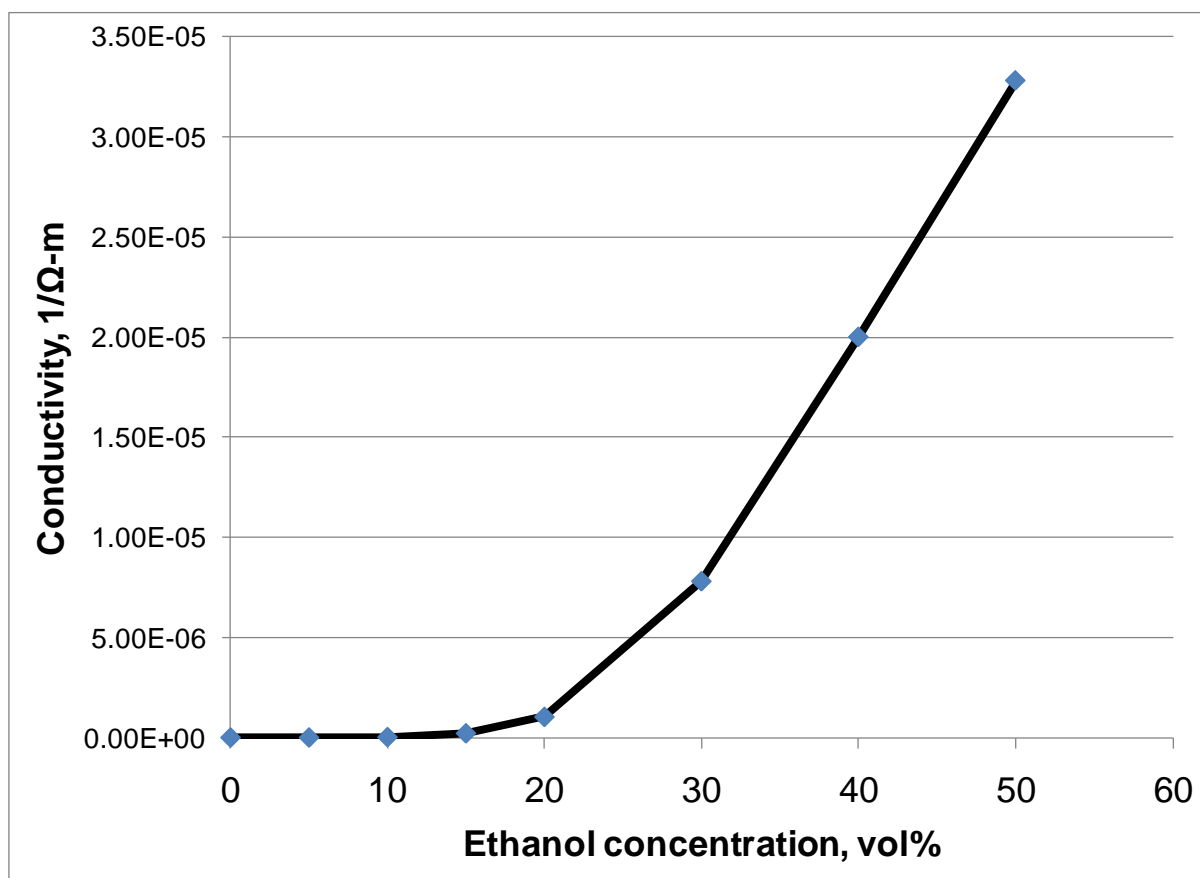
UST stakeholders generally consider fueling infrastructure materials designed for use with E0 to be adequate for use with E10, and there are no known instances of major leaks or failures directly attributable to ethanol use. It is conceivable that many compatibility issues, including accelerated corrosion, do arise and are corrected onsite and, therefore do not lead to a release. However, there is some concern that higher ethanol concentrations, such as E15 or E20, may be incompatible with current materials used in standard gasoline fueling hardware. In the summer of 2008, DOE recognized the need to assess the impact of intermediate blends of ethanol on the fueling infrastructure, specifically located at the fueling station. This includes the dispenser and hanging hardware, the underground storage tank, and associated piping.

The DOE program has been co-led and funded by the Office of the Biomass Program and Vehicle Technologies Program with technical expertise from the Oak Ridge National Laboratory (ORNL) and the National Renewable Energy Laboratory (NREL). The infrastructure material compatibility work has been supported through strong collaborations and testing at Underwriters Laboratories (UL). ORNL performed a compatibility study investigating the compatibility of fuel infrastructure materials to gasoline containing intermediate levels of ethanol. These results can be found in the ORNL report entitled *Intermediate Ethanol Blends Infrastructure Materials Compatibility Study: Elastomers, Metals and Sealants* (hereafter referred to as the ORNL intermediate blends material compatibility study).<sup>3</sup> These materials included elastomers, plastics, metals and sealants typically found in fuel dispenser infrastructure.

The test fuels evaluated in the ORNL study were SAE standard test fuel formulations used to assess material-fuel compatibility within a relatively short timeframe. Initially, these material studies included test fuels of Fuel C, CE10a, CE17a, and CE25a. The CE17a test fuel was selected to represent E15 since surveys have shown that the actual ethanol upper limit can be as high as 17%. Later, CE50a and CE85a test fuels were added to the investigation and these results are being compiled for a follow-on report to be published in 2012. Fuel C was used as the baseline reference and is a 50:50 blend of isooctane and toluene. This particular composition was used to represent premium-grade gasoline and was also used as the base fuel for the ethanol blends, where it is denoted by “C” in the fuel name. The level of ethanol is represented by the number following the letter E. Therefore a 10% blend of ethanol in Fuel C is written as CE10a, where “a” represents an aggressive formulation of the ethanol that contains water, NaCl, acetic and sulfuric acids per the SAE J1681 protocol.

## **1.2 ETHANOL COMPATIBILITY AND SOLUBILITY**

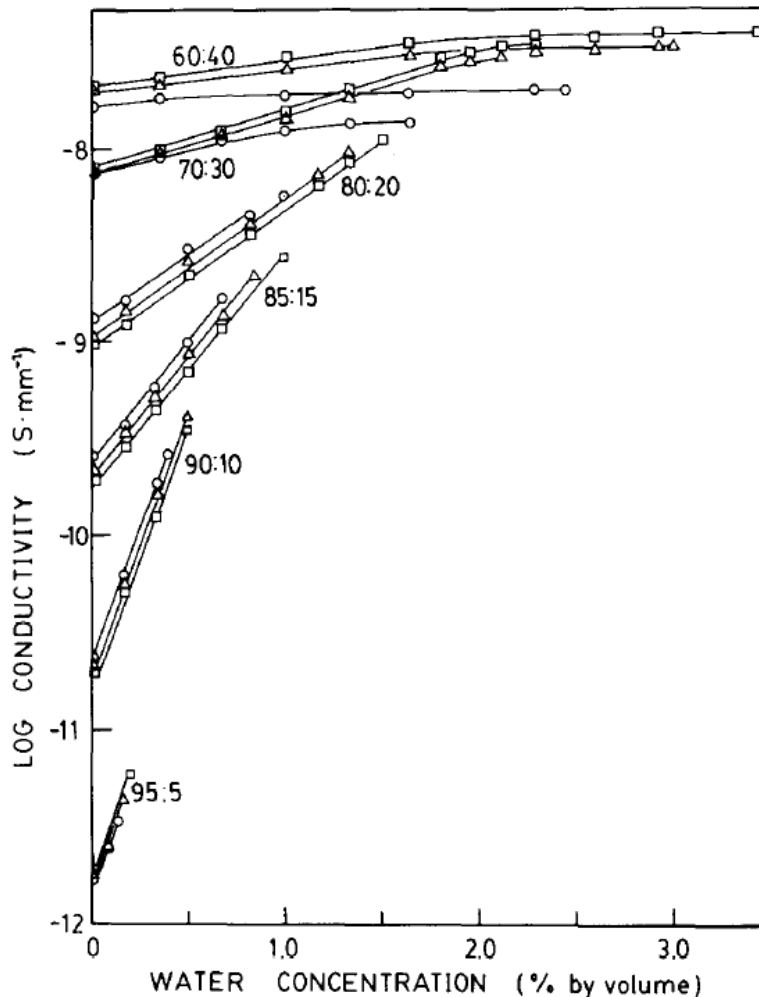
Pure ethanol, by itself, is not generally considered corrosive toward most metallic materials; however, as a polar molecule, ethanol will be more susceptible to having compatibility issues with both metals and polymers due to (1) increased polarity relative to gasoline, (2) adsorption of water, and (3) a higher solubility potential relative to gasoline. The first two factors are relevant to metals and alloys, while the latter affects primarily polymers. The corrosion potential is directly related to the electrical conductivity of a solution. Kirk<sup>10</sup> measured the electrical conductivity for gasoline as a function of ethanol concentration and dissolved water level. A plot of the electrical conductivity as a function of ethanol concentration in gasoline is shown in Fig. 1. As shown in the figure, the electrical conductivity is low for ethanol-blended gasoline increases marginally with ethanol concentrations up to 20%. However, although the conductivity numbers are low, relatively speaking, E15 is 10 times more conductive than E10. As the ethanol concentration increases from 20% to 50%, the corresponding conductivity increases by almost two orders of magnitude. As a result, metal corrosion becomes a significant concern for gasoline blends containing 50% or more ethanol.



**Fig. 1. Electrical conductivity of gasoline as a function of ethanol concentration.** *Source:* D. W. Kirk, Fuel 62, 1512–1513 (December 1983).

The level of dissolved water also has a pronounced effect. The results in Fig. 2 show the effect of water concentration in addition to ethanol level. In this figure, the electrical conductivity (listed as  $S$  in Fig. 2 and  $1/\Omega$  in Fig. 1) is plotted for blends containing 5, 10, 15, 20, 30 and 40% ethanol by volume. As the level of ethanol increases, the conductivity curves for each blend increase as well, and for each set of curves the conductivity also increases with the level of dissolved water. In fact, the water solubility limit increases the conductivity by an order of magnitude when going from E10 to E15. In addition, water itself is a solvent for NaCl and acids, which can lead to even higher rates of corrosion.

Ethanol also affects the material-fluid mutual solubility associated with the fuel blend, which is an important parameter for gauging the compatibility of fuels with polymers. The influence of the solubility parameter is complex; however, solvents and solutes having similar solubility parameters will have a greater affinity for permeation and dissolution.<sup>11</sup> The solubility parameter, or more specifically, the difference in parameters between the solute (polymer) and solvent (fuel), is important in predicting and understanding the solubility of a system. As the solubility parameter values for the solute and solvent converge, the propensity for the two components to mix (or allow the solvent to permeate into the solute) becomes thermodynamically possible. For an elastomer or plastic, this effect will be an increase in swelling of the polymer.



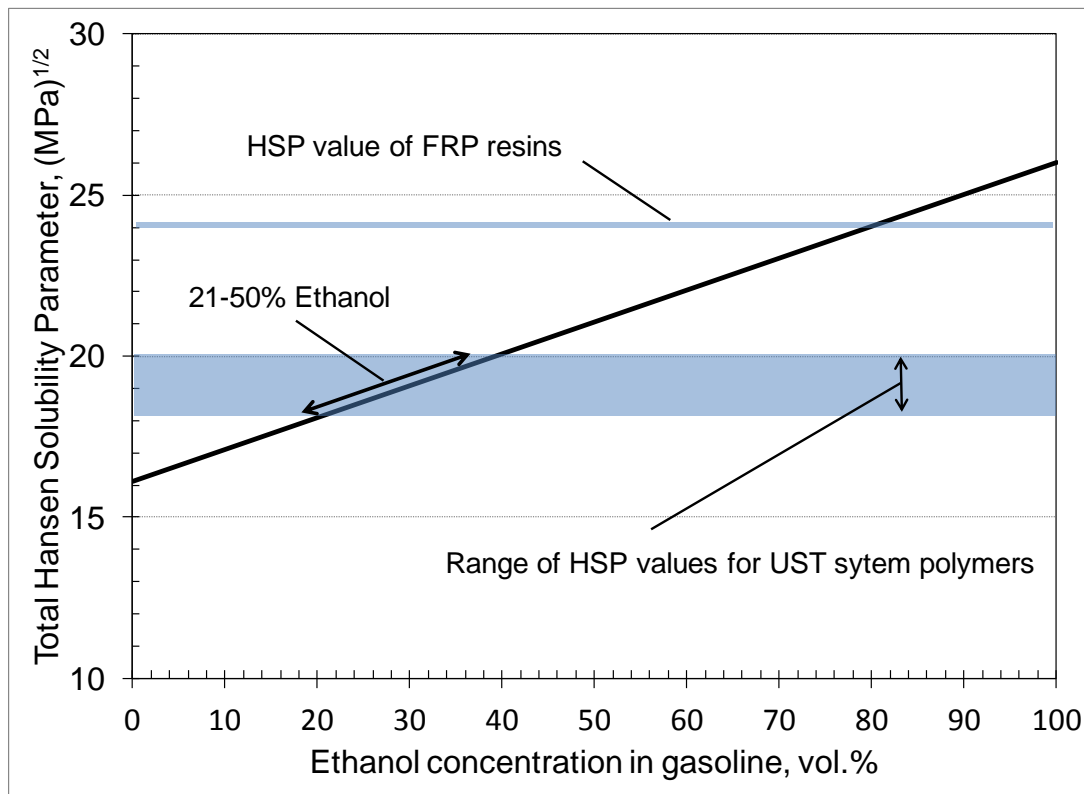
**Fig. 2. Electrical conductivity of gasoline as a function of ethanol and water content.** *Source:* D. W. Kirk, Fuel 62, 1512–1513 (December 1983).

A simplified representation of solubility as a function of ethanol concentration in gasoline is shown in Fig. 3. The wide shaded horizontal band in the chart represents the range of solubility parameters, expressed as total Hansen Solubility Parameter (HSP) for many dispenser polymers, especially elastomers. Epoxies, such as those used as the matrix materials for underground storage tanks, have a total HSP value around  $24(\text{MPa})^{1/2}$ , which is noticeably higher than the HSP for polymers. The implication for UST resins is that the solubility of the epoxy in the fuel will be highest for gasoline containing around 80% ethanol.

As the ethanol concentration increases from zero to 15%, it effectively raises the solubility parameter and approaches the solubility parameter of most dispenser polymers. Therefore, the propensity for the fuel to permeate into and dissolve polymeric components is enhanced. It is important to note that, in reality, solubility is determined from multiple thermodynamic factors, and that the highest level of mutual solubility for a given polymer does not necessarily match precisely with the theoretically-derived parameters which have been simplified in Fig. 3. Standard gasoline fuel delivery systems contain elastomeric materials having excellent compatibility and stability with hydrocarbon fuels. However, the ethanol molecule is relatively small and highly polar due to the  $-\text{OH}$  group. In addition the tendency to introduce hydrogen bonding is high. These features enable ethanol's permeation into and interaction with



the elastomer structure, which can result in swelling and softening of elastomers. Another negative feature associated with permeation is that soluble components, especially plasticizers added to impart flexibility and durability in the elastomer, may be leached out, thereby affecting the mechanical properties of the compounded elastomer component and degrading the ability of the component to perform its intended function.



**Fig. 3. Total Hansen Solubility Parameter as a function of ethanol concentration.** The lower blue horizontal band represents the solubility range of many UST system elastomer and plastics. The upper blue band is representative of FRP resins.

Several studies have been undertaken to evaluate the compatibility of ethanol with engine materials, especially those used in fuel system components such as pumps, and much of this work has recently focused on the intermediate E15, E20, and E25 blends.<sup>12-15</sup> However, little work had been reported on the compatibility of these fuels to standard fuel dispenser materials, which subsequently became the focus of the ORNL-led materials compatibility study noted earlier.

### 1.3 FUELING DISPENSER MATERIALS AND ETHANOL COMPATIBILITY STUDY

As part of the ORNL intermediate-blend materials compatibility study, an extensive survey was performed to identify to the extent possible all materials used in the fueling dispenser infrastructure. A list of the materials identified and evaluated in the ORNL study is shown in Table 1, where those materials identified by the authors of this report for use in UST systems are highlighted. Most of the plastic materials are used as structural components in FRP tanks and in both FRP and flexible piping systems. The elastomeric materials most identified as seals and gaskets are Viton™ and Dyneon™ brand fluorocarbons, but NBR and rubberized cork may still be in use in legacy tank probes and overfill devices. Steel is used in tanks and piping and aluminum is also used in some applications, such as drop tubes.

It is important to note that while the researchers were able to discover and identify an extensive list of relevant materials over the course of this and other studies, it is possible, if not probable, that other materials used in legacy, and some new infrastructure systems, were not included in this investigation.

**Table 1. List of materials evaluated in intermediate ethanol blends compatibility study. (Materials identified as being used in UST systems are highlighted.)**

Metals/Alloys	Elastomers	Plastics	Sealants
<b>304 stainless steel</b>	<b>Viton™ fluorocarbon</b>	<b>High density polyethylene (HDPE)</b>	<b>PTFE-based sealants (two-types) with and without Teflon™ tape</b>
<b>1020 carbon steel</b>	<b>Dyneon™ fluorocarbon</b>	Fluorinated HDPE	
<b>1100 aluminum</b>	<b>Acrylonitrile butadiene rubber (NBR)</b>	Polypropylene (PP)	
Cartridge brass	Silicone rubber	Polyoxymethylene	
Phosphor bronze	Silicone rubber	<b>Nylon</b>	
Nickel 201	Fluorosilicone rubber	<b>Polyvinylidene fluoride (PVDF)</b>	
Terne-plated steel	Neoprene rubber	<b>Polytetrafluoroethylene (PTFE)</b>	
Galvanized steel	Styrene butadiene rubber (SBR)	Polyphenylene sulfide (PPS)	
Cr-plated brass	Polyurethane	<b>Polyethylene terephthalate (PET)</b>	
Cr-plated steel	<b>Rubberized cork</b>	Polybutylene terephthalate (PBT)	
Ni-plated		Polythiourea	
Ni-plated steel		<b>Isophthalic ester resin</b>	
		<b>Terephthalic ester resin</b>	
		<b>Vinyl ester resin</b>	
		<b>Epoxy resin</b>	

Of the all the test fuels investigated (Fuel C, CE10a, CE17a, CE25a, CE50a and CE85a), only the metal and elastomeric materials were subjected to each fuel type. The plastics were originally exposed to Fuel C and CE25a (and later to CE50a and CE85a) and the sealants were evaluated only in Fuel C, CE10a and CE25a. At a later point in this study, ORNL received sections of fiberglass USTs removed from use. Three UST sections were cut into test specimens and added to the final exposure runs of Fuel C, CE50a, and CE85a.

The test protocol consisted of immersing the specimen coupons in the test fuels and vapors for extended periods, 4 weeks for metals and elastomers and 16 weeks for plastics. During the exposure period the fuel temperature was maintained at 60°C in order to maintain consistency with the UL Subject 87A-E25 test standard used in by Underwriter Laboratories when assessing fuel compatibility.<sup>16</sup>

## 2. UNDERGROUND TANKS & PIPING SYSTEMS

Underground fuel storage tanks are composed either of steel or fiberglass reinforced plastic. Both of these materials, as well as flexible plastic, are also used in piping systems. A breakdown of the piping types using an analysis based on 22 state databases<sup>5</sup> is shown in Table 2. The overwhelming majority of installed piping (~71%) is either flexible or rigid fiberglass reinforced plastic. Of the remaining metal systems, approximately 18% of metal piping systems are steel. Copper makes around 2% of underground piping and approximately 8% is of unknown material construction.<sup>19</sup> The most common installed piping systems are rigid FRP and flexible plastic systems. Older piping systems were typically single-walled, but most newly installed systems are double-walled. FRP makes up approximately 58% of installed piping, while flexible plastic piping accounts for around 13% of all installed piping systems.<sup>5</sup>

**Table 2. Breakdown of piping materials.**<sup>5,19</sup>

Material Class	Approximate Percentage Used as of 2009
Steel	18
Rigid Fiberglass Reinforced Plastic (RFP)	58
Flexible Plastic Piping	13
Other (copper, PVC, etc.)	2
Unknown	8

A large percentage of leaks occur in the piping system between the tank and the dispenser.<sup>17</sup> These leaks typically occur at joints and connections where the stresses are highest. Contributors to stress include movement and forces exerted on piping from environmental factors which can be caused by changes in ground-water level and settling changes in the soil. Even a small change in the position of a UST will result in stress on the piping, especially at joints.<sup>18</sup> The level of stress will be higher for rigidly designed systems as opposed to flexible systems which can reduce stress through bending and relaxation. Outside of environmental contributions to stress, there are inherent changes caused by the piping materials' response to the fuel chemistry. As stated earlier, ethanol will raise the solubility parameter of the fuel so that the resulting potential for degradation of plastics is increased. Increased solubility will likely cause an increase in the volume of the plastic. This volume increase will place the component pipe under additional elongation and stress. Expansion of piping caused by solubility (even at low levels of approximately 2%) may be high enough to lead to failure based on life cycle studies of polymeric piping materials.<sup>18</sup>

## **2.1 METALLIC MATERIALS FOR TANKS, COMPONENTS, AND PIPING SYSTEMS**

Steel is commonly used as a tank material for both legacy and newer systems, and steel piping is estimated to be used in approximately 18% of piping systems. The other metallic material that is exposed directly to E10 (and potentially E15) is aluminum which is used in submersible pumps. Both steel (carbon and stainless) and aluminum were included in the ORNL intermediate-blend materials compatibility study.

As shown in Fig. 1, the electrical conductivity for E10 and E15 is low in relationship to higher ethanol concentrations; however, when compared to each other, E15 is actually 10 times more conductive than E10. When water is added to levels approaching the solubility limit (as shown in Fig. 2), the conductivity is further increased. The test fuels used in the ORNL-intermediate blends study included relatively high levels of dissolved water (0.09% of the total ethanol volume) to account for this factor. In this study, steel and aluminum, along with the other metal coupons (tested either as single components or galvanic couples) showed negligible corrosion from exposure to the test fuels.<sup>20</sup> As a result, corrosion that does occur on metal tanks or piping systems is likely due to one of more of the following factors (none of which were included in the ORNL study):

1. Phase separation of water from the ethanol fuel blend
2. External water intrusion from rain, humidity, etc.
3. Contamination by other means such as road salt, dirt, etc.
4. Stress corrosion cracking

The potential for aqueous phase separation can be discussed relative to Fig. 2. As shown in Fig. 2, the level of water that can be dissolved into E15 is roughly twice the amount that can be dissolved in E10. The higher water content translates to a higher potential for corrosion.

## 2.2 POLYMER PIPING AND TANK SYSTEMS

As shown in Table 2, the majority of underground piping is constructed from plastic materials, which are categorized as two types, flexible piping and FRP piping. Although FRP systems are more established in the field, the majority of new piping systems installed today are flexible plastic systems because these systems are easier to install. As a result, the percentage of flexible piping is expected to grow relative to the other piping systems over the next 10 years. The piping arrangement can consist of either single- or double-walled systems. The majority of installed single-walled piping systems are legacy units, but new requirements are resulting in increased use of double-walled piping systems. Double-walled systems have an interstitial space between the walls that can be monitored for leaks.

### 2.2.1 Flexible Plastic Piping

Typical compositional arrangement of flexible piping includes an inner barrier liner within a layer of fiber reinforcement (to provide strength) and a cover to protect the inner layers from damage from handling and to prevent water intrusion. We surveyed the materials used in the construction of the outer wall for double wall plastic-based systems. In virtually every case, the outer wall is composed of inexpensive materials, known to be less chemically resistant to ethanol.

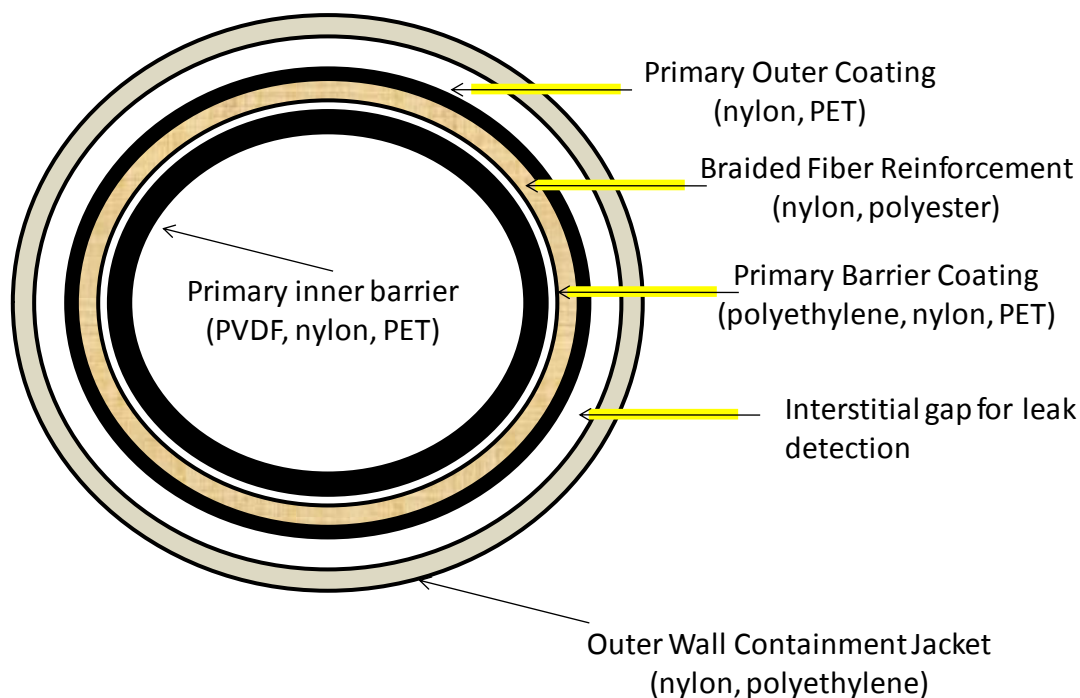
Multiple piping manufacturers and the materials used in their systems are listed in Table 3.<sup>21-25</sup> Some of the manufacturer and material information included in the table was taken from surveys dating to 1997, and therefore, may not reflect current construction.

**Table 3. Flexible piping materials according to manufacturer.**

<b>Manufacturer</b>	<b>Permeation Barrier Material</b>	<b>Reinforcement</b>	<b>Primary Pipe Cover Material (Single-walled)</b>	<b>Secondary Containment Materials (double-walled)</b>
Advanced Polymer Technology	Nylon 12	Nylon fiber wrap	Polyethylene	HDPE
Ameron	PVDF	Polyester braid polyethylene	Nylon	HDPE
Containment Technologies	Selar nylon (amorphous)	None	Polyethylene	HDPE
Environ	PVDF	Polyester braid	Nylon-coated polyethylene	Nylon coated polyethylene
Furon	PVDF	Polyester braid	Nylon II	
OPW	PVDF	Polyester braid	Nylon II	
PetroTechnik	Nylon	None	Polyethylene	Polyethylene
Total Containment	Carilon polyketone (product discontinued)	Polyester or Kevlar braid	Polyethylene	Polyethylene
Western Fiberglass	PVDF	Polyester braid	Nylon II	HDPE
XP-Piping	Nylon 12 and mylar (PET)	Nylon fiber	Mylar (PET) coated nylon 12	Nylon 12
Pisces	Kynar (PVDF)	Nylon fiber	Nylon	Nylon
Geoflex	Kynar (PVDF)		Nylon coated polyethylene	Nylon coated polyethylene

Of the flexible pipes reported in Table 3, the majority had inner barrier layers composed of PVDF. The three remaining designs incorporated nylon, either as nylon 12, Selar™ amorphous nylon, or a combination of nylon 12 and Mylar™ PET. For most systems the permeation barrier layer was externally reinforced with wound fibers composed of either nylon or polyester. This reinforcement, in turn, is usually coated with nylon or polyethylene. Likewise, the most common materials used for the outer wall are polyethylene and nylon.

One manufacturer used PET as the inner barrier layer. However, most materials are either nylon or PVDF. In reality the actual arrangement and location material arrangement for flexible piping is somewhat complex. A cutaway diagram showing the material arrangement for one commercially-available flexible pipe is shown in Fig. 4. In all flexible piping systems, there is an inner permeation barrier layer composed of a plastic material that has low solubility (i.e., high resistance) to petroleum fuels and alcohols.



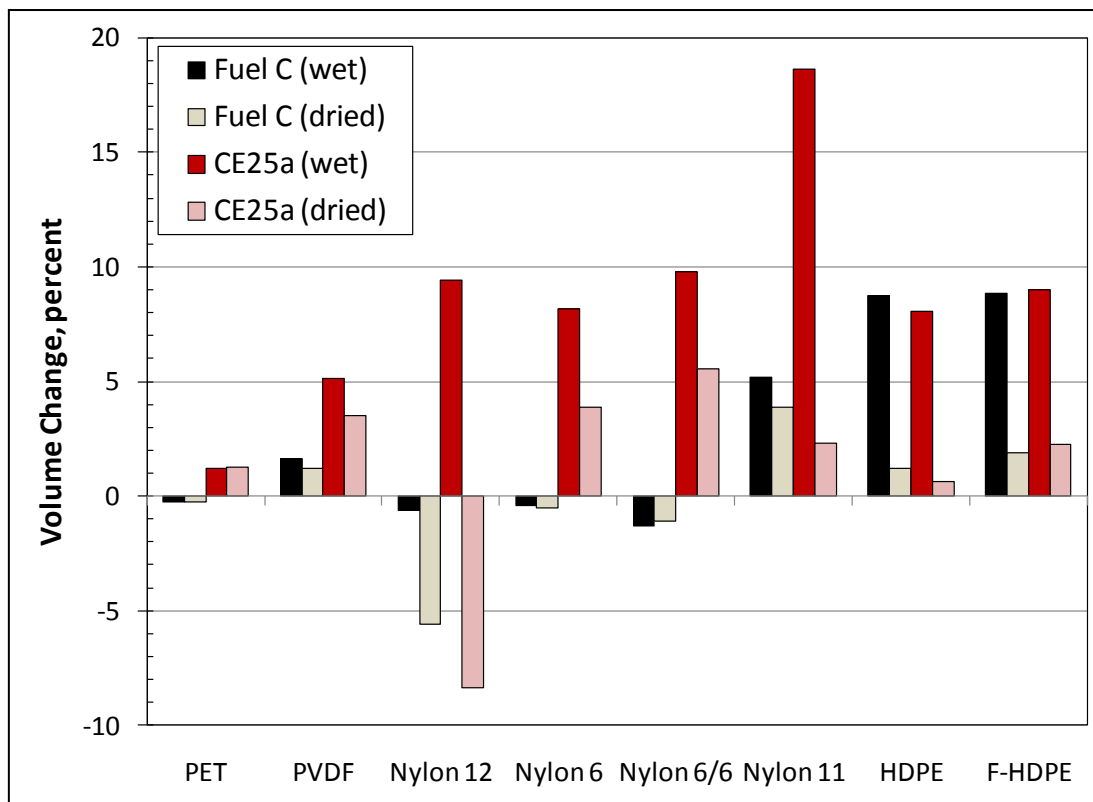
**Fig. 4. Cross-section diagram of flexible piping showing an example of the layering position and arrangement of materials used in double-walled designs.** A typical single-wall design is similar but would not include the outer wall containment jacket shown on the outside.

As stated earlier the two primary polymer types used in flexible fuel piping are nylon and polyvinylidene difluoride (PVDF). Other often used materials are polyketone and polyethylene terephthalate (PET). However, polyketone (Carilon™, Dupont) was discontinued and (to the best of our knowledge) the installed piping was removed and replaced. PET is more expensive than either nylon or PVDF, and as such, is not extensively used in piping applications. PVDF goes by the tradename, Kynar™ and is manufactured by Arkema, Inc. The other established material is the DuPont Selar™ nylon barrier material (which is amorphous grade of nylon).

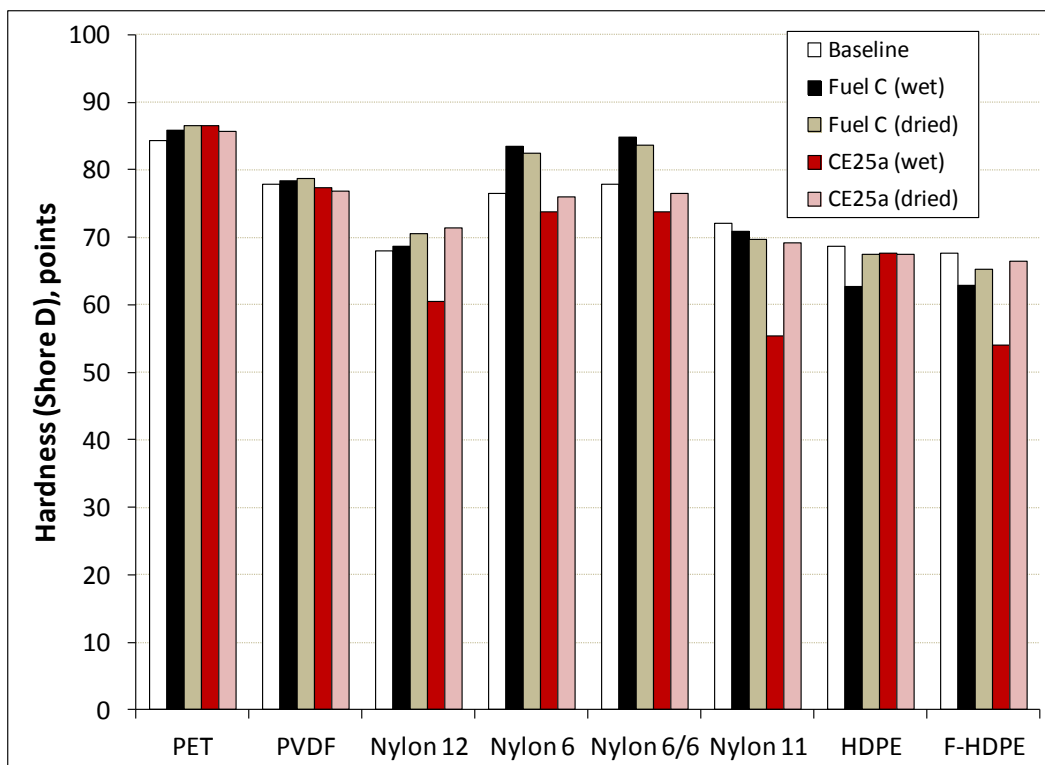
Flex piping is easier to install and the flexible nature of the material allows the component to relax during swell. In contrast to fixed rigid piping systems, a flexible piping system can undergo small dimensional changes in volume and movement (relaxation), thereby reducing the stress load.

The ORNL intermediate blends compatibility study included samples of representative flexible pipe materials. These materials include PET, HDPE, nylon 6, nylon 6/6, nylon 11, and nylon 12. (Selar, which is an amorphous grade of nylon, was not evaluated.) These nylon grades are differentiated by the degree of molecular alignment (crystallinity), additives, and processing. In contrast to the other types, Nylon 11 is a unique specialty grade made from vegetable oil. Although Selar™ nylon was not specifically included among the test coupons, according to DuPont, its chemical resistance is comparable to other grades of synthetic nylon (nylon 6, 6/6, and 12).<sup>26</sup>

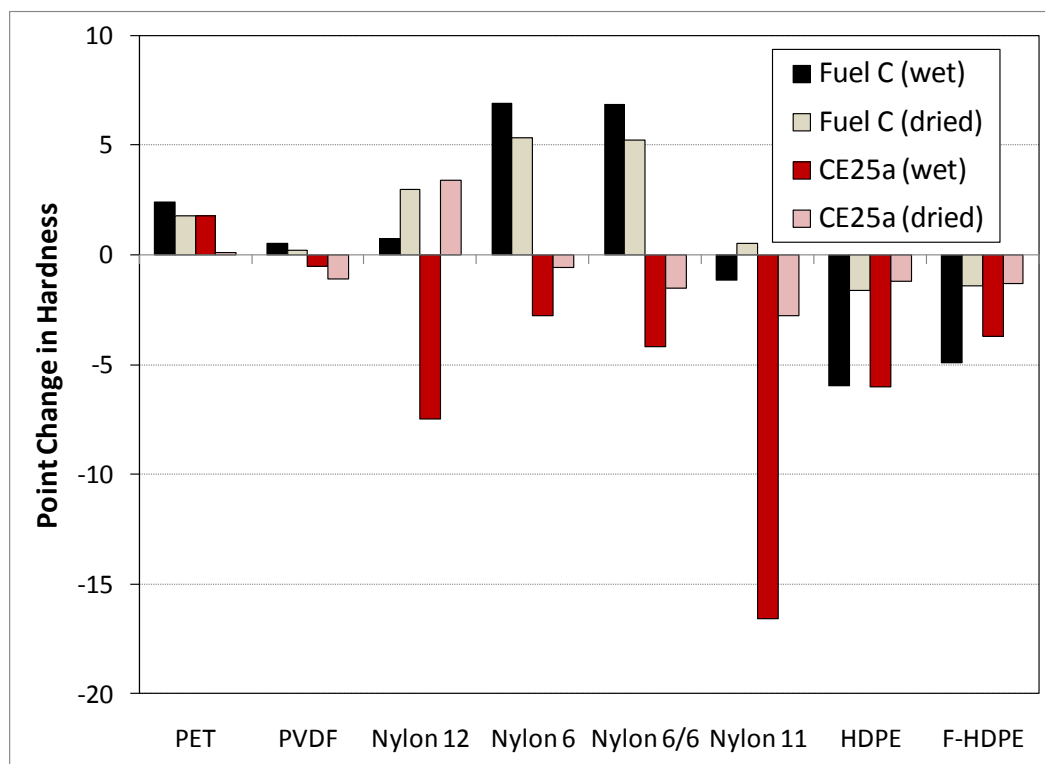
The ORNL materials compatibility study evaluated the response of selected plastic materials to Fuel C and CE25a only. Test fuels representing 10 and 15 percent aggressive ethanol were not exposed to plastics. Volume swell and hardness results are shown in Figs. 5 through 7 for common nylon grades, PET, PVDF, and HDPE exposed to Fuel C and CE25a.



**Fig. 5. Volume swell results for representative barrier materials used in underground piping.**



**Fig. 6. Absolute hardness results for representative barrier materials used in flexible piping.**



**Fig. 7. Point change in hardness (from baseline) for representative barrier materials used in flexible piping.**

As shown in Fig. 5, PET and PVDF experienced the lowest low volume swell (1.23% and 5.12%, respectively) following exposure to CE25a. In contrast, the nylon 6, nylon 6/6, and nylon 12, along with the HDPE samples swelled between 8 and 10 %. The highest level of volume swell occurred for nylon 11, and was around 18 %. Following dry out, these materials retained some fluid, as evidenced by the residual swell present in the dried samples. The one exception is nylon 12, which shrank from its original volume to over 5% from exposure to Fuel C and around 8% with CE25a. Such shrinkage is evidence that Fuel C and CE25a were able to dissolve and remove a significant portion of the solid material. The hardness results presented in Figs. 6 and 7 show that nylon 12 and nylon 11 both became softer with exposure to CE25. The decrease in hardness of nylon 12 was around 7 points, which is only marginally higher than the softening of the nylon 6, nylon 6/6 and the HDPE samples. However, nylon 11 dropped 17 points and this drop coupled with the high volume swell suggests that nylon 11 may not be acceptable for use in plastic piping, even for E0 formulations.

Although E10 and E15 test fuels were not evaluated, an estimation of the volume swell can be made using solubility parameters (obtained from the literature) and volume swell results in CE25a. Volume swell is a measurement of solubility. According to solubility theory, the difference between the solubility parameters is inversely related to the solubility between the solute (plastic) and solvent (test fuel). In other words, the closer match between the total Hansen Solubility Parameters of the solute and solvent, the more mutually soluble they are to each other. Using the known total HSP values for the plastic materials and E25, E15 and E10, and the measured volume swell in CE25a (as shown in Table 4), a calculated volume swell for each material in E15 and E10 can be made using the ratio of the differences in the total Hansen solubility parameters between the plastic and CE25a to the HSP difference between the plastic and CE15 and CE10. These calculated values are shown in Table 5.

The method for calculation of volume swell is as follows:

$$VS_{(EX)} = VS_{(E25)} (1 - (\Delta HSP_{(EX)} - \Delta HSP_{(E25)})) / (\Delta HSP_{(EX)})$$

Where:

$VS_{(EX)}$  is the volume swell of the plastic sample after exposure to a fuel containing X percent of ethanol by volume.

$VS_{(E25)}$  is the volume swell of the plastic in CE25a

$\Delta HSP_{(EX)}$  is the difference between the total Hansen Solubility Parameter values for the plastic and the fuel containing X volume percent ethanol

$\Delta HSP_{(25)}$  is the difference between the total Hansen Solubility Parameter values for the plastic and the fuel containing 25 volume percent ethanol



**Table 4. Volume swell results for representative barrier materials used in flexible underground piping**

Plastic	Hansen Solubility Parameter (MPa <sup>1/2</sup> )	Volume Swell in CE25a (%)
PVDF	23.17	5.12
Nylon 6	20.3	8.15
Nylon 12	22.2	9.40
PET	20.8	1.23
Fuel Type	Hansen Solubility Parameter (MPa <sup>1/2</sup> )	
E25	18.58	
E15	17.59	
E10	17.09	

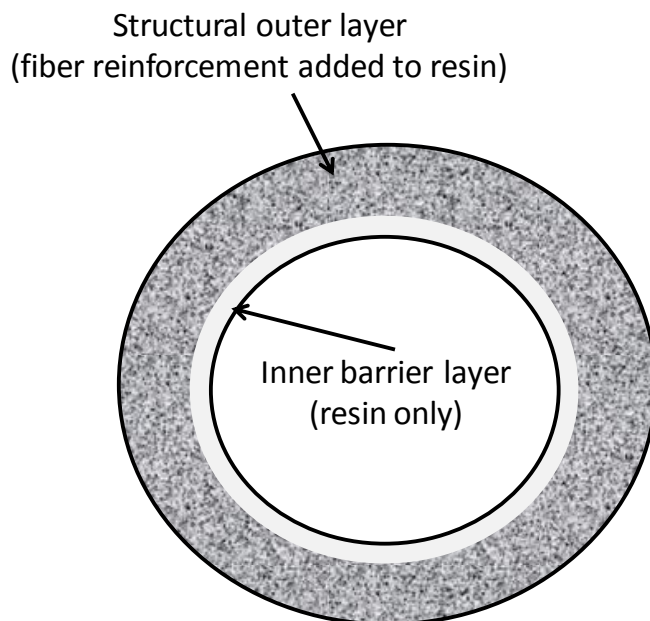
**Table 5. Measured and calculated results for PVDF and Nylon 6**

Barrier material	HSP (MPa <sup>1/2</sup> )	Measured volume swell (CE25a)	Calc. Volume Swell for E15	Calc. Volume Swell for E10	Estimated vol. increase associated with increasing ethanol from E10 to E15
PVDF	23.17	5.12	4.1	3.6	0.5
Nylon 6	20.3	8.15	5.2	4.4	0.8
Nylon 12	22.2	9.4	7.4	6.7	0.7
PET	20.8	1.23	0.8	0.7	0.1

The results in Table 5 show that the expected increase in volume swell when going from E10 to E15 is less than 1 percent for the primary barrier liner materials used in flexible piping. The low additional volume swell is not likely to create much stress in the piping since these materials are able to relax due to the flexible nature of the piping. Based on these results, we do not anticipate any noticeable potential for release associated with going from E10 to E15. However, if the piping is rigidly constrained somehow, then stress buildup may occur to cause bucking (or cracking) of the piping. Most of the changes in swell (and hardness) will occur from moving from E0 to E10.

### 2.2.2 Fiber-reinforced Plastic Tanks & Piping

Fiber-reinforced plastic piping materials, design and construction are similar to those used in fiberglass tanks. The construction consists of first placing resin on a mandrel and later adding fiber reinforced resin to serve as the outer layer. A diagram showing layering and arrangement is depicted in Fig. 8.



**Fig. 8. Diagram of fiber-reinforced plastic piping.**

As shown in Fig. 8, the inner barrier liner is approximately 0.5mm-thick resin layer surrounded by a much thicker (~6mm) layer of fiber-reinforced resin. For FRP systems used to contain petroleum fluids, the fiber reinforcing material is fiberglass.

ORNL tested several FRP resins to assess compatibility with CE25a. The resins that were evaluated included:

1. Isophthalic polyester resin (1 part isophthalic acid to 1 part polyester resin) known as Vipel F701- This resin type was used extensively in USTs prior to the 1990s.
2. Isophthalic polyester resin (2 parts isophthalic acid to 1 part polyester resin) known as Vipel F764- This resin type was used in USTs starting in the 1990s.
3. Terephthalic polyester resin (2 parts terephthalic acid to 1 part polyester resin) known as Vipel F774- This resin type was used extensively in 1990s.
4. Epoxy novolac vinyl ester resin known as Vipel F105- This is the most recently advanced corrosion resistant UST resin.

A survey of manufacturers shows that these resins are the most commonly used types for FRP UST construction.<sup>27-30</sup> It is important to note that for FRP tanks the construction does not consist of a multilayer structure similar to the arrangement used in flexible plastic piping. The inner barrier layer consists solely of the resin material, with fiberglass added to the thicker resin outer layer to provide strength and elasticity. The volume swell results are shown in Fig. 9 for these resins. (These specimens consisted of pure resin and did not contain fiber reinforcement.) Vipel F701 swelled to over 15 volume percent upon exposure to Fuel C. However, these samples fractured during dry-out making it impossible to ascertain accurate volume swell. For Vipels F764 and F774, the volume swell in Fuel C was around 9 percent and 7 percent, respectively. The most compatible grade was Vipel F085, which exhibited low volume swell (2%). When dried at 60°C for 20 hours, the volume swell was lower than the wetted condition, but still significantly higher than the starting condition. The increase in dry-out volume

compared to the initial condition indicates that significant levels of Fuel C are contained within the resin. This fact is further illustrated in Fig. 10 which shows the corresponding mass change of the specimens before and after dry-out.

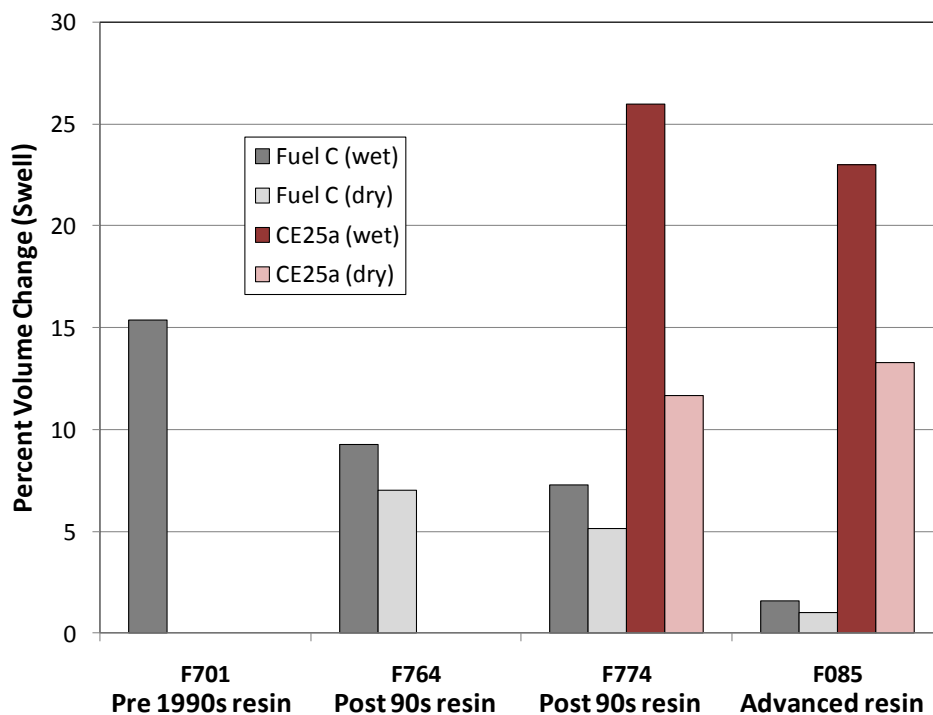


Fig. 9. Volume swell results for UST resins following exposure to Fuel C and CE25a.

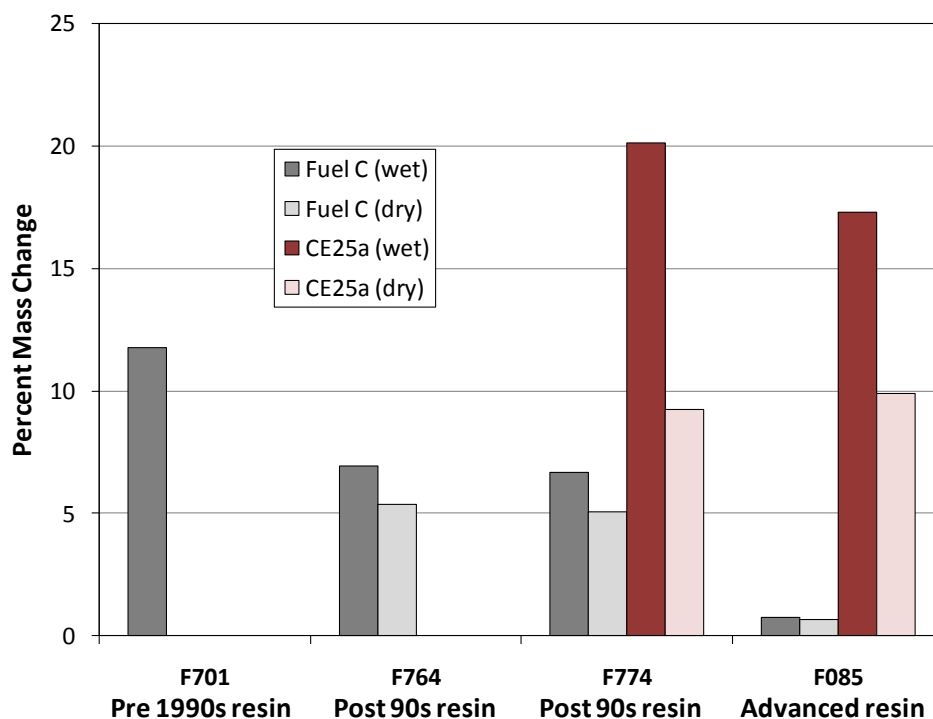


Fig. 10. Mass change for UST resins following exposure to Fuel C and CE25a.

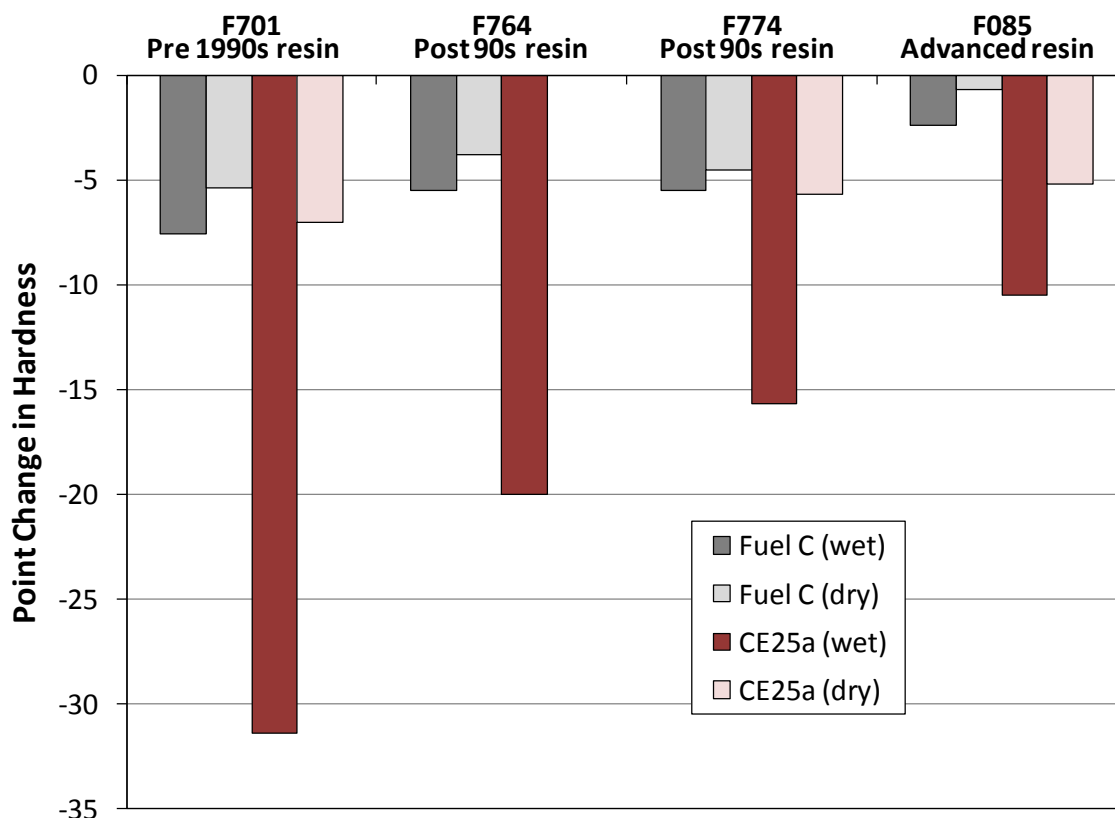
Upon exposure to CE25a, Vipels F774 and F085 exhibited high degrees of swelling. F774 swelled to 26% while F085 swelled to around 23%. However, because the resins are reinforced with glass fibers, the actual swelling of the composite structure will be considerably lower. The inner barrier liner will be more susceptible to expansion, but the fiberglass reinforcement will prevent outward expansion of the resin barrier layer. However, inward expansion may occur and this effect may result in softening or cracking, or other forms of damage. It is important to note that the specimens which cracked in the test fuels (Vipel F701 and F764) were composed of pure resin, and these resins are not designed for use without fiberglass reinforcement. Fiberglass, by itself, is insoluble and is used in composite structures to provide modulus and strength. Fiberglass reinforcement would resist fuel permeation and elongation in the composite structure. As a result the level of swell in FRP systems would be expected to be much lower than for the pure resin. However, the inner barrier layer of an FRP tank (or FRP piping) is not reinforced and may experience degradation in the form of softening, spalling, or cracking.

The estimated volume swell for Vipel F774 and Vipel F085 in E10 and E15 was calculated using the Hansen Solubility Parameter-based method employed for estimating the volume swell for the flexible plastic materials. These results are shown in Table 6, and show that for the Vipel F774 and F085 resins, the difference in calculated volume swell associated with E10 and E15 is low (around 1.5% for both resin types). This means that in all likelihood there will be minimal effect when moving from E10 to E15. However, there is potential for a big difference when moving from E0 to E10.

**Table 6. Measured and calculated results for UST resins Vipel F774 and Vipel F085**

<b>Resin material</b>	<b>HSP (MPa<sup>1/2</sup>)</b>	<b>Measured volume swell (CE25a)</b>	<b>Calc. Volume Swell for E15</b>	<b>Calc. Volume Swell for E10a</b>	<b>Estimated vol. increase associated with increasing ethanol from E10 to E15</b>
Vipel F774	24.1	25.99	22.0	20.5	1.5
Vipel F085	24.1	22.99	19.5	18.1	1.4

The change in hardness results for the UST resins are shown in Fig. 11. For each resin type, the hardness dropped slightly with exposure to Fuel C, but CE25a was shown to significantly lower hardness in the wetted condition. Vipel F701 and Vipel F764 exhibited greatest drop hardness (31 and 20 points, respectively) from the original condition. These values are considered high and since hardness is a measure of strength and elastic modulus, it is not surprising that these two specimens exhibited fracture following exposure to CE25a. Interestingly, Vipel F774 also experienced a relatively large decrease in hardness of around 15 points. The combination of reduced volume swell and lower change in hardness (relative to the F701 and F764 resins) were enough to prevent fracture of the F774 resin. The most advanced resin grade, Vipel F085 exhibited the least change in hardness of the resins tested.



**Fig. 11. Point change in hardness for the UST resin samples following exposure to Fuel C and CE25a.**

Vipul F701 was used extensively in fiberglass reinforced USTs prior to 1990.<sup>25</sup> As such, it was designed primarily for gasoline use only and was not optimized for compatibility with ethanol-blended fuel. The volume swell and hardness decrease upon exposure to Fuel C would be considered acceptable for this resin type. F701 was replaced with more ethanol-resistant grades during the 1990s. Any legacy tanks composed of F701, or similar resin type, may be subject to ethanol degradation. Although the 30-year warranty on tanks composed of F701 would have expired by now, many of these tanks are still in use. During the 1990s, many of the isophthalic resins were replaced by terephthalic-based and epoxy vinyl resins for improved performance.<sup>28</sup> The data provided by the materials compatibility testing shows that the terephthalic and vinyl resins are better suited for ethanol compatibility.

The ORNL intermediate-blend study also evaluated coupons taken from FRP underground storage tanks that had been removed from service. These tanks were cut into sections and sent to ORNL for evaluation. Photographs showing these specimens before and after exposure to the test fuels are shown in Figs. 12, 13, and 14. In each figure the baseline represents the unexposed sample.

Unfortunately the resin formulation of the UST sections was unknown, but the tanks were most likely pre-1990s vintage. Therefore, the resin formulations used in these tanks may not have been designed for use with ethanol-blended fuel. Over one dozen tanks were sectioned and sent to ORNL. All, except one, of these USTs were of amber coloration, similar to the pure resin coupons that were discussed previously, and nearly identical in appearance. There was one set of tank sections that was unique in that it was the only UST to have a corrugated plastic film adhered to the inner surface and was dark green in coloration. Test coupons were cut from three UST sections, which were labeled Batch 1, Batch 2, and Batch 3. Both

Batch 1 and Batch 2 were identical in appearance and of amber coloration, while Batch 3 was taken from the green section, which also contained the plastic film. Batch 3 was chosen since it represented an arrangement and coloration different from the rest.

Three coupons from each UST were evaluated in Fuel C, CE50a and CE85a test fuels. (The UST sections were not included in the earlier CE10a, CE17a, or CE25a test fluids, since this activity was started after these studies were completed.) These coupons were exposed in the test fluids for 16 weeks at 60°C along with other plastic specimens.



Fig. 12. Photograph showing the Batch 1 specimens before and after exposure to Fuel C, CE50a, and CE85a.

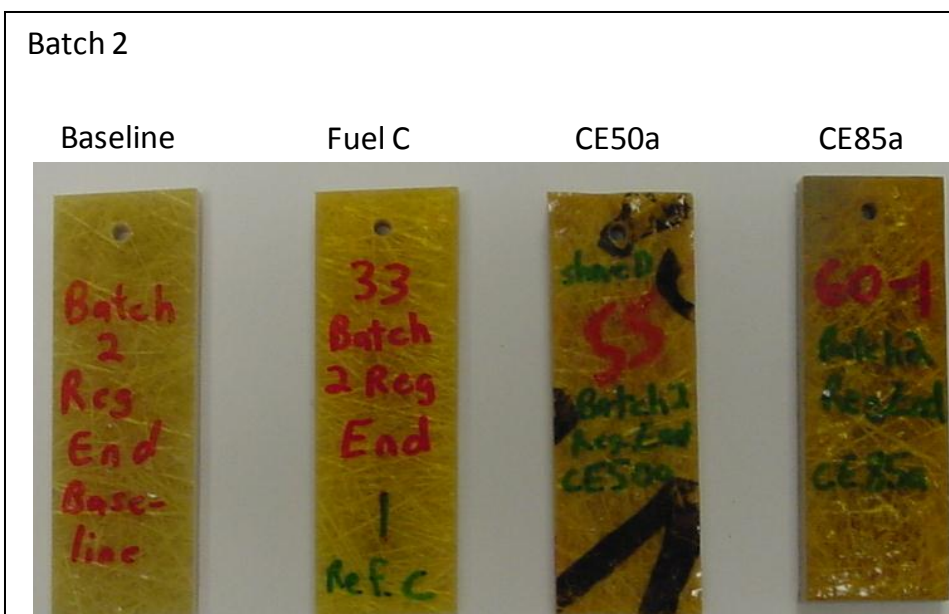
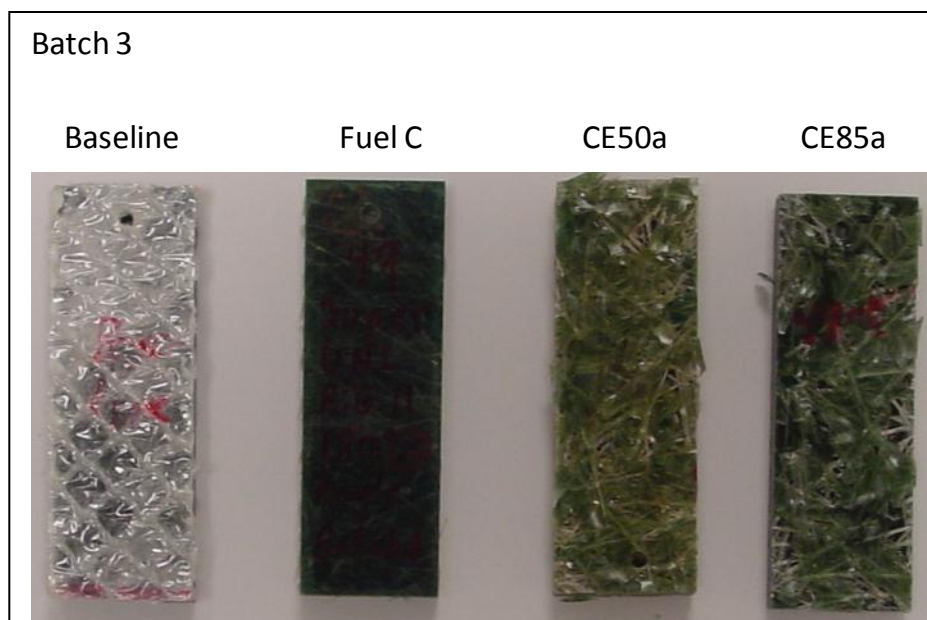


Fig. 13. Photograph showing the Batch 2 specimens before and after exposure to Fuel C, CE50a, and CE85a.



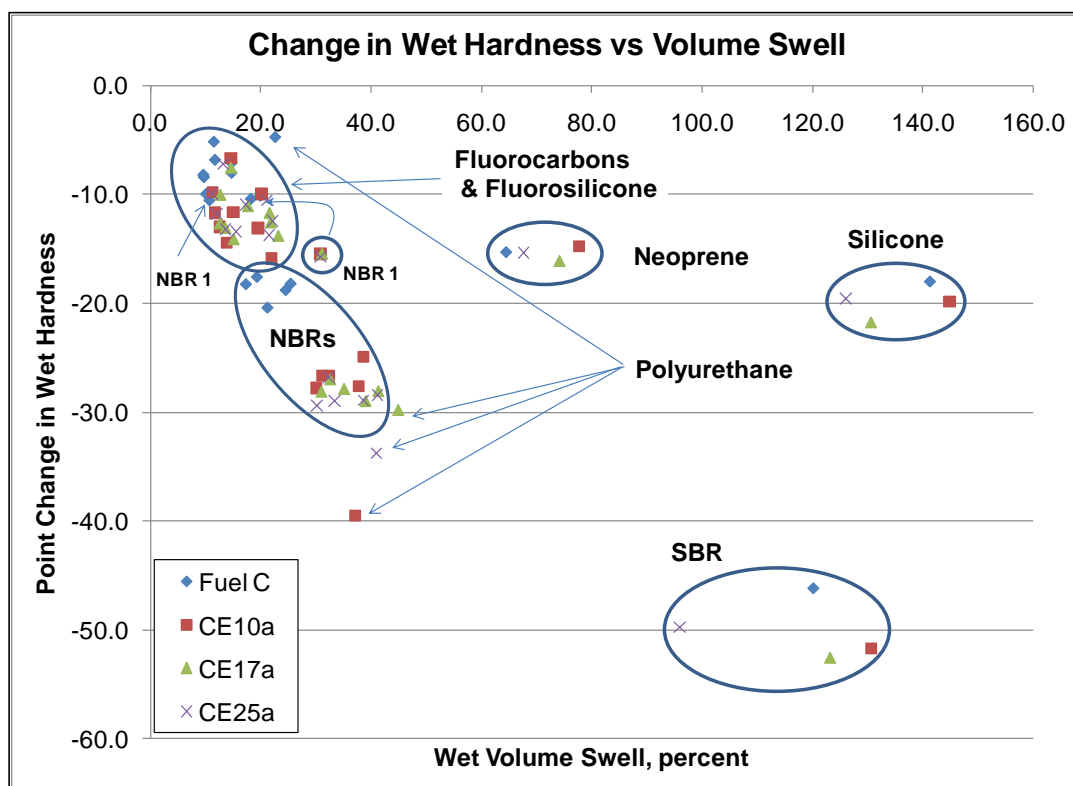
**Fig. 14. Photograph showing the Batch 3 specimens before and after exposure to Fuel C, CE50a, and CE85a.**

The photographs shown in Figs. 12 and 13 reveal that the amber resin specimens (Batch 1 and Batch 2) did not experience any observable degradation (outside of a slight change in color) from exposure to ethanol. However, the Batch 3 specimens (shown in Fig. 14) experienced massive degradation from the CE50a and CE85a test fuels. For this design, the corrugated liner was debonded by the Fuel C and the aggressive ethanol fuels. Interestingly, this liner survived exposure to the test fuels. However, the inner resin layer was removed and the resin surrounding the fiberglass reinforcement had dissolved to the extent that the fibers were completely exposed. It is important to note that, as depicted in Fig. 3, epoxy-based resins are likely to be more soluble in CE50a and CE85a fuels than for intermediate E10 and E15 levels. Therefore it is expected that the Batch 1 and Batch 2 USTs will be compatible to gasoline containing intermediate levels of ethanol. However, if the corrugated liner of the Batch 3 UST was damaged or breached, then it is likely that this UST has a high risk of leaking.

### **3. ELASTOMERS, SEALANTS, COUPLINGS AND FITTINGS**

#### **3.1 ELASTOMERS**

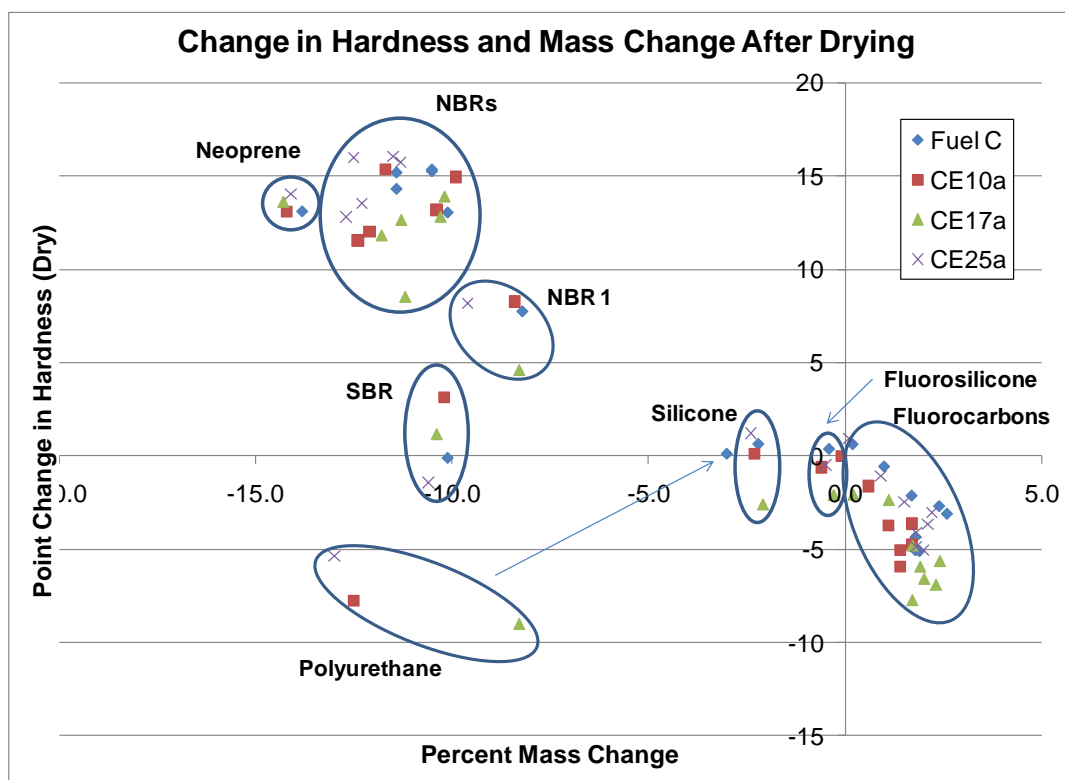
Although elastomers are ubiquitous in fuel dispenser components, especially as hoses and seals, they are not used extensively as primary piping materials in either FRP or flexible piping systems. However, these elastomers could be used as gaskets and seals in the submersible tank pump head. A survey of piping and coupling manufacturers listed Viton fluorocarbon as the only o-ring material sold today for use in couplings and fittings for gasoline delivery systems.<sup>31</sup> Other elastomer types were not mentioned as coupling materials for current and new UST piping systems, although they may be prevalent in legacy systems. Of the elastomers evaluated in the ORNL intermediate-blend materials compatibility study, fluorocarbons were found to be the most compatible to ethanol. The other elastomers, in particular nitrile rubbers (NBRs), showed moderate but significant increases (10-12%) in swell and increased softening with exposure to aggressive ethanol as shown in Fig. 15. However, the additional increase associated with CE17a exposure (compared to CE10a) was small (5 to 8%).



**Fig. 15. Volume swell and point change in hardness for elastomers exposed to Fuel C, CE10a, CE17a and CE25a.**

During dry-out, elastomers such as NBR and neoprene exhibit moderate shrinkage and embrittlement (see Fig. 16) which is attributed to extraction of the plasticizer components. However the level of shrinkage or mass reduction associated is constant and independent of ethanol content. As a result, the increase in leak potential among the elastomers when moving from E10 to E15 is expected to be low. However, these materials (especially NBR, neoprene, and SBR) will exhibit a high increase in swell when moving from E0 to E10 (or E15). Therefore, care must be taken when placing ethanol-blended gasoline into a system that had only contained gasoline.





**Fig. 16. Percent mass change and point change in hardness for elastomers exposed to Fuel C, CE10a, CE17a and CE25a following dry-out.**

It is important to note that these elastomers are used solely as seals (i.e., o-rings, gaskets, etc.) and are not utilized as structural materials for UST systems. Additional swell for o-rings and gaskets in some cases does not degrade seals or diminish sealing potential, and may, to a small degree, improve the performance of the seal. These materials are not recommended for use as structural components of piping and UST systems, since even moderate levels of swell will create internal stresses which, even at low levels, can significantly reduce the lifecycle and durability of a component.

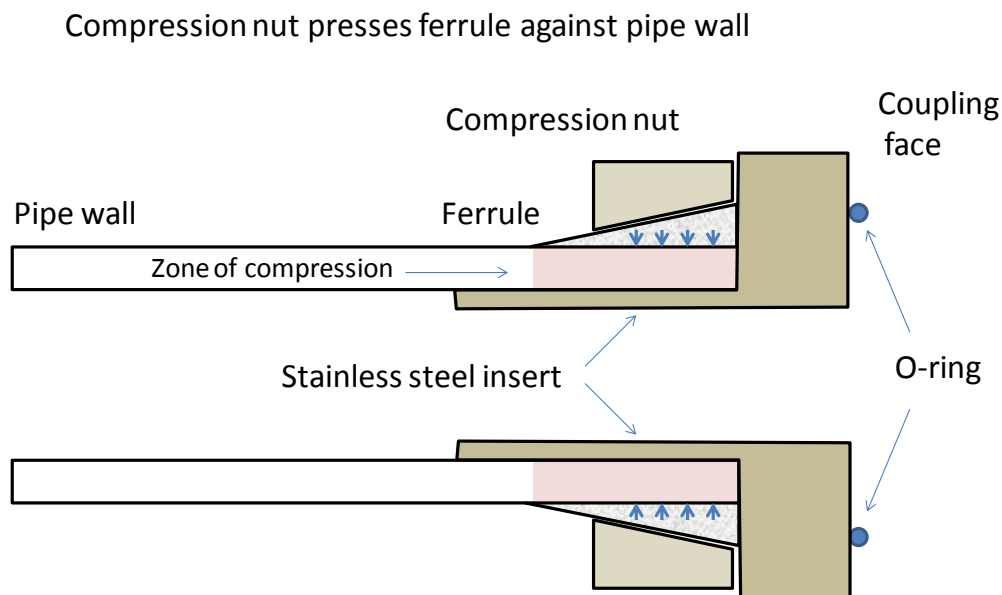
### 3.2 PIPE THREAD SEALANTS

Pipe thread sealants are used for metal piping and some FRP piping systems. Standard PTFE sealants (such as RectorSeal™) were originally developed for E0 use and were used extensively in legacy piping systems. These sealants have been shown to be incompatible for use with alcohols. In the ORNL intermediate-blend materials compatibility study, RectorSeal™ was shown to be incompatible with CE10a. This result strongly indicates that the pipe thread sealants used in the E0 legacy systems experienced leaking when exposed to E10. Ethanol compatible sealants such as GasOila ESeal™ were subsequently developed for ethanol-blended gasoline use and are now the industry standard. The ORNL study showed that the GasOila ESeal™ product is compatible with fuel containing up to 25 percent aggressive ethanol. It is very likely that the standard PTFE sealants used in the legacy systems were replaced with the ethanol-compatible products during the implementation of E10. There is no hard data to support this assessment, but based on the development and widespread use of the GasOila product, it appears to be the case. Except for polyurethane (which is used as a coating rather than as a seal), the elastomers and sealants evaluated in the ORNL intermediate-blend materials study showed no significant increase in swell and softening when moving from E10 to E15. Therefore, we do not foresee any added potential for releases when switching from E10 to E15.

### 3.3 COUPLINGS AND FITTINGS

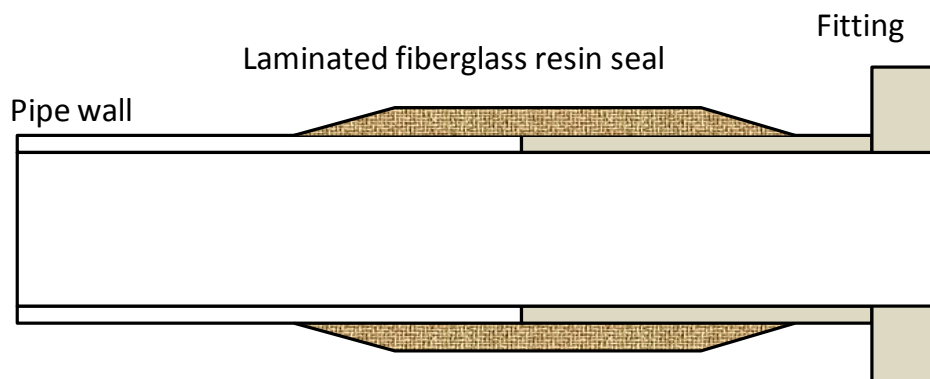
One of the most susceptible locations of the underground storage tank systems are the couplings used to connect piping, fittings, and valves. There are two potential sources of leaks. One is where the coupling attaches to the piping and the other one is at the seal interface mating two couplings together. The interfacial seal issue was discussed under the elastomer section and is not considered to be a significant point of release if seals and gaskets are made of fluorocarbon materials and are properly installed.

Flexible plastic piping typically utilizes swage-type fittings to join piping and connect valves and flanges. A typical coupling assembly consists of a stainless steel insert with one or two o-rings, a stainless steel ferrule with one o-ring, and a swivel nut (or other means) to compress the ferrule against the outer pipe surface.<sup>31</sup> A simplified schematic is shown in Fig. 17. The compression of the plastic between the stainless steel insert and ferrule maintains a leak tight seal. In this configuration, the fuel is only exposed to the plastic piping, stainless steel coupling and the o-ring used to seal the coupling adjacent faces. Newer units were found to utilize fluorocarbon as the o-ring material, although legacy couplings may use other elastomers (such as NBR). These couplings usually require a special tool (from the piping supplier) to install properly. It is important to note that couples for FRP piping cannot be installed in this manner because the hard resin would fracture under high compression.



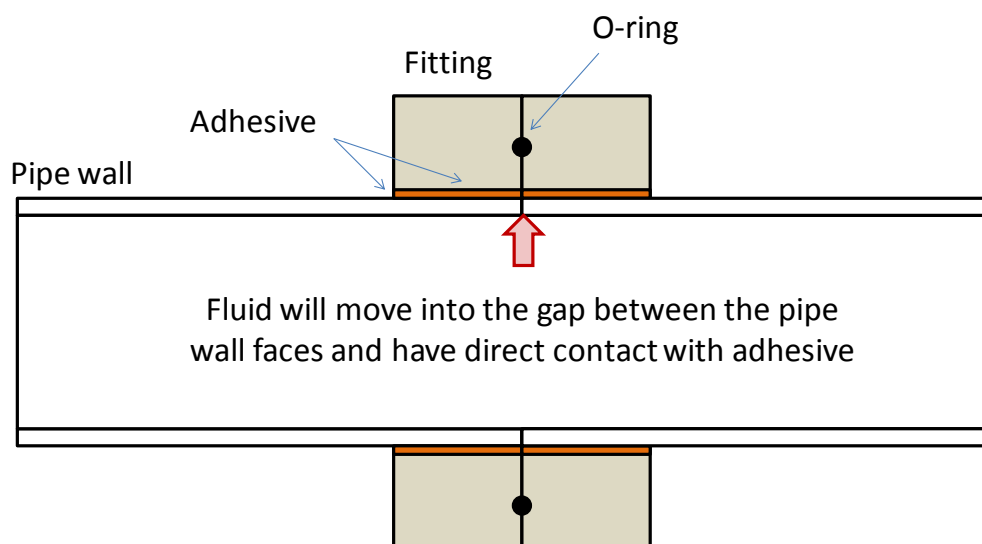
**Fig. 17. Simplified schematic showing attachment of a coupling to flexible plastic piping.**

The two most common methods for joining FRP piping and attaching couplings are adhesive bonding and butt and strap joints.<sup>32</sup> The butt and strap method is considered the most reliable means for joining FRP piping. Two pieces of pipe are butted together and layers of chopped fiberglass are wrapped around the pipe in a resin matrix similar to the pipe composition. A diagram depicting a butt and strap joint is shown in Fig. 18. If the butt and strap materials are similar or identical to the pipe materials, then compatibility performance is expected to be essentially the same and thus potential for further degradation due to E15 is minimal.



**Fig. 18. Schematic diagram of the butt and strap joint.**

The adhesive method involves adding an adhesive to glue a fitting or coupling to FRP. Because of its inherent weakness, this method is not used to join pipe sections, but is restricted to attaching fittings (such as flanges). Typically the outer surface of the pipe is sanded to allow better distribution of the sealant and to enable the adhesive to better grip the pipe surface and thereby form a strong mechanical bond. One application of this method is shown in Fig. 19. The adhesive maintains a seal between the fitting and the pipe end. At the fitting face, the adhesive will be exposed to fuel in the crevice region between the adjacent pipe ends. For some applications, the outer pipe walls are tapered at the ends to enable better fit between the pipe wall and fitting.



**Fig. 19. Schematic showing a common arrangement of using adhesive on FRP piping.**

The adhesives used for FRP systems contain a mix of inorganics (over 50%), such as clay, limestone, and silica and a mix of hydrocarbons.<sup>33</sup> The inorganic fraction imparts strength, rigidity and is resistant to attack from aggressive fuel components (including alcohols). The remaining hydrocarbons consist primarily of acrylic polymers, resins and distillate products. Information pertaining to adhesive resistance to ethanol was lacking and we were not able to ascertain ethanol compatibility.

## 4. CONCLUSIONS

The USEPA Office of Underground Storage Tanks commissioned a study at ORNL to evaluate whether an increased potential for leaking of USTs will occur when moving from E10 to E15 fuel. The original intention was to construct a probabilistic failure analysis tool to estimate the increase in releases, if any, if E15 replaced E10 in regulated UST systems. A key part of this process was to solicit opinions from a panel of industry and regulatory experts to identify critical variables that impact failure likelihood estimates. However, the lack of information on the performance of existing UST systems with E15 precluded the possibility that state/industry experts could speculate on E15's impact to UST systems. Therefore, the project objective was redirected to address the added leak potential (or incompatibility) of UST system materials when moving from E10 to E15. The data used to make this assessment were obtained primarily from the ORNL intermediate blend compatibility study. This study included metal and polymeric materials typically used in UST systems, and these materials were evaluated in aggressive test fuel formulations representing E0, E10, E15, E25, E50 and E85. Potential leak locations, such as pipe couplings were identified, and the elastomers and sealants used in couplings and joining were also studied.

### 4.1 CONCLUSION ON TANKS AND PIPING MATERIALS

Metallic materials included carbon and stainless steel and aluminum. A large number of USTs are composed of carbon steel, which is also used in approximately 18% of piping. Stainless steel is used in pipe couplings which are used to join piping sections and fittings. Aluminum, while not used as extensively as either carbon or stainless steels, is used in the construction of submersible pumps. However, failure of a submersible pump should not lead to leaking. The results from the ORNL intermediate-blends compatibility study showed that carbon and stainless steels, and aluminum will not undergo significant corrosion in either E10 or E15. However, it is important to note that the test conditions for these materials did not include stress or water-phase separation, both of which can contribute to increased corrosivity. In fact, if aqueous phase separation occurs, then the risk for corrosion will be higher for E15 since the maximum level of dissolved water is roughly twice that of E10.

Plastics are used extensively in underground piping systems. The two types of plastic piping, flexible and FRP, employ different types and grades of plastic materials. Flexible piping is primarily composed of various grades of nylon, PVDF, PET, polyester, and polyethylene. These materials were only tested in Fuel C and CE25a. As a result, the volume change, associated with CE10a and CE15a exposure, was estimated using the known swelling behavior at CE25a and Hansen Solubility Parameters for the plastics and test fuels. Nylon 11 exhibited the highest level of swelling (~18%) and would likely not be considered acceptable for use in USTs or flexible piping systems. Likewise, nylon 12 also may not be acceptable due to the significant loss of mass after drying. Other plastics, such as HDPE, F-HDPE, nylon 6 and nylon 6/6 exhibited relatively high swell (8-10%) and may not be suitable when switching from E0 to either E10 or E15. However, the calculated swell for nylon 6, nylon 6/6, PVDF, PET, and polyethylene indicated that the added increase in swell when moving from E10 to E15 was very low. This result suggests that the leak potential in E15 for flexible piping containing these materials will be low as well.

The performance of resins used in the construction of FRP tanks and piping is highly dependent on the type of resin. A pre-1990 legacy isophthalic polyester resin was visibly damaged with exposure to a test fuel containing 25% aggressive ethanol. Analysis of post-1990s resins (exposed to CE25a) were mixed; the resin composed of isophthalic polyester was damaged, while the resins composed of terephthalic polyester or vinyl ester were not. Interestingly, the two resins that were damaged from exposure to ethanol were both isophthalic polyesters. Based on these results, isophthalic polyester resins should be avoided in the construction of UST systems storing ethanol-blended fuels. The predicted level of volume

swell associated with E10 and E15 was calculated for the terephthalic polyester and vinyl ester resins. The results suggest that the added volume swell associated with E15 (compared to E10) is extremely low and would not likely increase the potential for leaking with E15 fuel. ORNL was able to include three legacy UST samples in a later compatibility effort using CE50a and CE85a as test fuels. In one unique case, a legacy FRP UST that contained a separate plastic liner exhibited significant degradation of the resin material when exposed to high levels of ethanol. Although the liner was not visibly damaged, its performance with lower intermediate levels of ethanol-blended gasoline could not be ascertained. The other two UST sections were not damaged and would likely exhibit good compatibility with E10 or E15.

## **4.2 CONCLUSION ON ELASTOMERS, SEALANTS, COUPLINGS AND FITTINGS**

A high leak potential also exists where piping sections are joined and fittings are attached. The structural material typically used in these applications is stainless steel and the sealing materials are either elastomers and/or pipe thread sealants. Modern joining units employ primarily fluorocarbons in o-ring and sealing applications; however some legacy systems may use NBRs and other elastomer types. The ORNL intermediate-blend ethanol compatibility study investigated the performance of fluorocarbons, fluorosilicone, NBRs, silicone rubber, styrene butadiene rubber, neoprene and polyurethane. These elastomers all showed significant swelling with exposure to ethanol. However, because elastomers are used solely as seals (i.e., o-rings, gaskets, etc.), swelling is not necessarily an indication of leak potential. Additional swell for o-rings and gaskets may improve the performance of the seal. Except for polyurethane (which is used as a coating rather than as a seal), the elastomers and sealants evaluated in the ORNL intermediate-blend materials study showed no significant increase in swell and softening when moving from E10 to E15. Therefore, for field applications and materials examined in this study, there should not be any corresponding potential for releases associated with increase the ethanol concentration in fuel gasoline from E10 to E15. The flanges used in coupling systems are composed of stainless steel and this material has been shown to have excellent compatibility with ethanol-blended fuels.

Pipe thread sealants are used for metal piping and some FRP piping systems. Standard PTFE sealants (such as RectorSeal™), used in E0 applications, were shown to be incompatible for use with E10. However, ethanol-compatible sealants (such as GasOila ESeal™) were compatible with fuel containing up to 25 percent aggressive ethanol. Although it is very likely that standard PTFE sealants used in legacy systems were replaced with the ethanol-compatible products during the implementation of E10, there may be systems still in use with the incompatible sealant material.

FRP piping joined using either a butt and strap configuration or an adhesive is used to secure a fitting on one end. The butt and strap consists of a FRP wrap that contains resin similar or identical to the FRP pipe resin, and therefore, should be compatible with ethanol-blended fuel. Adhesives consist of a mix of various organic and inorganic materials, and we could not assess their compatibility to ethanol since they were not included in the ORNL intermediate-ethanol blends compatibility study.

In general, several materials evaluated in this study were found to not perform well in fuel blends containing ethanol. These materials demonstrated incompatibility with E10 and should not be used for E15 (unless it can be demonstrated that a particular polymer grade is, in fact, compatible). Systems most susceptible to increased leakage will be those legacy USTs which are currently using E0 and will be switching directly to E15.

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